

# Lab Report: Acid and Base Investigation

## Hypothesis (5 pts)

My hypothesis was that a substance was acidic, indicating a pH value less than 7. I expected to have an orange color.

Substance: vinegar  
 Indicator: litmus  
 Expected pH: acidic  
 Expected Color: orange  
 Actual Color: orange

## Procedure

1. Select a substance that you want to test and determine that based on its acid nature.
2. Prepare a small test tube. Monitor the color of the test tube as the acid reacts.
3. Transfer it to the test tube rack.
4. Add some acid (vinegar) to the test tube and observe its color change.
5. Using the test tube, determine the acid nature of the substance (acidic or basic).
6. Add the substance to the test tube and observe its color change.
7. Record the color change in the table. Determine the pH value of the substance. It is an acidic or basic or neutral substance.
8. Repeat the experiment.

## Data (5 pts)

	Substance to be tested	Indicator	Color change	Result
Substance	<u>vinegar</u>	<u>litmus</u>	<u>orange</u>	<u>acidic</u>
Indicator	<u>litmus</u>	<u>litmus</u>	<u>orange</u>	<u>acidic</u>
Expected pH	<u>acidic</u>	<u>acidic</u>	<u>orange</u>	<u>acidic</u>
Expected Color	<u>orange</u>	<u>orange</u>	<u>orange</u>	<u>orange</u>
Actual Color	<u>orange</u>	<u>orange</u>	<u>orange</u>	<u>orange</u>

## Conclusion (5 pts)

1. The substance was acidic, as indicated by the color change from orange to red.
2. The substance was acidic, as indicated by the color change from orange to red.
3. The substance was acidic, as indicated by the color change from orange to red.
4. The substance was acidic, as indicated by the color change from orange to red.

# Acid And Base Investigation Lab Report

**SA Dillow**



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