

LAB: ACID BASE TITRATION



Questions

1. How accurate was your titration? What was your percent error?

2. What was

Name _____

Titration Lab

Objective: Students will be able to determine the concentration of an unknown solution by performing a titration.

Background: A titration is a technique where a solution of known concentration is used to determine the concentration of an unknown solution. Typically, the titrant (the known solution) is added from a buret to a known quantity of the analyte (the unknown solution) until the reaction is complete. A color change indicates that the solution has reached the equivalence point. The color change occurs because an indicator that illustrates the equivalence point of acid and base has been added to the flask. For the purposes of this experiment, the equivalence point will be when the solution is neutralized.

Materials:

NaOH Buret Clamp
HCl Buret

Ring Stand
Graduated Cylinders

Flask
Indicator

White paper (optional)

Safety Information:

Hydrochloric Acid is a skin and eye irritant. Always wear appropriate goggles when handling this acid. Sodium hydroxide is a strong base. It should not come in contact with your skin. If sodium hydroxide gets on your skin, wash the area with plenty of water.

Procedure:

1. Set up the titration apparatus, as shown to the top right.
2. Measure out unknown acid and place it into a flask. Record amount added.
3. Add 2 drops of indicator. (Your teacher may do this step)
4. Pour NaOH into the buret using the funnel - be careful not to spill.
5. Begin to titrate the NaOH into the Acid. This is only to be a few drops at a time.
6. MAKE SURE you

Acid Base Titrations Chemfax Lab 6 Answers

James Sherwood Fritz



Acid Base Titrations Chemfax Lab 6 Answers:

This book delves into Acid Base Titrations Chemfax Lab 6 Answers. Acid Base Titrations Chemfax Lab 6 Answers is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Acid Base Titrations Chemfax Lab 6 Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Acid Base Titrations Chemfax Lab 6 Answers
- Chapter 2: Essential Elements of Acid Base Titrations Chemfax Lab 6 Answers
- Chapter 3: Acid Base Titrations Chemfax Lab 6 Answers in Everyday Life
- Chapter 4: Acid Base Titrations Chemfax Lab 6 Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Acid Base Titrations Chemfax Lab 6 Answers. This chapter will explore what Acid Base Titrations Chemfax Lab 6 Answers is, why Acid Base Titrations Chemfax Lab 6 Answers is vital, and how to effectively learn about Acid Base Titrations Chemfax Lab 6 Answers.
3. In chapter 2, this book will delve into the foundational concepts of Acid Base Titrations Chemfax Lab 6 Answers. This chapter will elucidate the essential principles that need to be understood to grasp Acid Base Titrations Chemfax Lab 6 Answers in its entirety.
4. In chapter 3, this book will examine the practical applications of Acid Base Titrations Chemfax Lab 6 Answers in daily life. This chapter will showcase real-world examples of how Acid Base Titrations Chemfax Lab 6 Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Acid Base Titrations Chemfax Lab 6 Answers in specific contexts. The fourth chapter will explore how Acid Base Titrations Chemfax Lab 6 Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Acid Base Titrations Chemfax Lab 6 Answers. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Acid Base Titrations Chemfax Lab 6 Answers.

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