Exp 13 Volumetric Analysis: Acid-Base titration

Exp. 13 - video (time: 47:17 minutes)

<u>Titration</u> - is the measurement of the volume of a standard solution required to completely react with a measured volume or mass of substance being analyzed (also referred to as analyte)

Standard solution - is a solution containing a known concentration of solute.

<u>Standardization</u> - the concentration of a solution is determined by using it to titrate a carefully measured quantity of a primary standard.

<u>Primary standard</u> - is a highly purified compound that serves as a reference material in all volumetric and mass titrimetric methods. (high purity, stable toward air/dried, absence of hydrated waters, cost, soluble in solvent, large molar mass)

Acid Base Titration Volumetric Analysis Lab Answers

Marco Cascella

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