Acid and Base Worksheet - Answers

- Write equations for the following acid-base reactions and indicate each conjugate acid-base pair:
 - a) HNO₃ + OH⁻ → H₂O + NO₃⁻ A B CA CB conjugate acid-base pairs: HNO₃ and NO₃⁻ OH⁻ and H₂O
 - b) OH⁻ + HPO₄⁻² → H₂O + PO₄³⁻
 B A CA CB

 conjugate acid-base pairs:
 HPO₄²⁻ and PO₄³⁻
 OH⁻ and H₂O
 - c) HSO₃" + HF → H₂SO₃ + F" B A CA CB conjugate acid-base pairs: HSO₃" and H₂SO₃ HF and F"
- Write the names for the following acids and bases:
 - a) KOH potassium hydroxide
 - b) H₂Se hydroselenic acid
 - c) C₂H₃O₂H acetic acid
- 3) Write the formulas for the following chemical compounds
 - a) ammonium sulfate (NH₄)₂SO₄
 - b) cobalt (III) nitride CoN
 - c) carbon disulfide CS₂

Acid Base Revew With Answers

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "Acid Base Revew With Answers," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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