Acid and Base Worksheet - Answers

- Write equations for the following acid-base reactions and indicate each conjugate acid-base pair:
 - a) HNO₃ + OH⁻ → H₂O + NO₃⁻ A B CA CB conjugate acid-base pairs: HNO₃ and NO₃⁻ OH⁻ and H₂O
 - b) OH⁻ + HPO₄⁻² → H₂O + PO₄³⁻
 B A CA CB

 conjugate acid-base pairs:
 HPO₄²⁻ and PO₄³⁻
 OH⁻ and H₂O
 - c) HSO₃" + HF → H₂SO₃ + F" B A CA CB conjugate acid-base pairs: HSO₃" and H₂SO₃ HF and F"
- Write the names for the following acids and bases:
 - a) KOH potassium hydroxide
 - b) H₂Se hydroselenic acid
 - c) C₂H₃O₂H acetic acid
- 3) Write the formulas for the following chemical compounds
 - a) ammonium sulfate (NH₄)₂SO₄
 - b) cobalt (III) nitride CoN
 - c) carbon disulfide CS₂

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