

- Identify dangers of caustic materials and potentially explosive reactions
- Describe evidence of chemical change in combustion, corrosion, and other oxidation reactions
- Infer evidence of a chemical reaction between household materials
- Infer whether a reaction is exothermic or endothermic based on observed or given evidence
- List ways to alter the rate of a chemical reaction
- Explain that mass is conserved during chemical reactions
- Identify patterns in the organization of the periodic table, especially the first 18 elements
- Explain the difference between observation and theory, and how observations are used to form theories
- Use the periodic table to identify the number of protons, electrons, neutrons in any given element on the periodic table
- List examples of ionic and molecular compounds and distinguish between them
- Read and interpret chemical formulas for compounds of two elements, and give the IUPAC name and common name of such compounds
- Identify household chemical compounds and write the chemical symbols
- Predict combining ratios in ionic compounds of two elements
- Identify the number of atoms in given molecules of molecular compounds
- Build and draw models of ionic and molecular compounds
- Describe chemical reactions using word equations or chemical equations
- Match IUPAC symbols to their meaning

Unit C: Environmental Chemistry

Students must be able to:

- Recall all their definitions
- Give examples of common organic and inorganic substances that are essential to living things, and explain why they're needed
- Describe proteins, carbohydrates, and lipids
- Explain how chemicals are introduced into the environment
- Describe how living things ingest or absorb materials
- Explain how some substances (DDT, heavy metals) are difficult for organisms to break down or get rid of
- Use environmental and scientific perspectives to decide what substances in what amounts can be safely added to the environment, and how
- List examples of substrates and nutrient sources for living things in various environments
- Explain how biological monitoring can indicate environmental quality
- Explain how chemical factors (like pH, dissolved oxygen or nutrients) affect the health and abundance of living things
- Calculate chemical concentrations in ppm, ppb, or ppt
- Use indicators and the pH scale to identify acids, bases, and neutral substances
- Describe the effects of acids and bases on each other and other substances AND living things
- Describe how materials are transported through the air, water, and soil, and how this can be sped up or slowed
- Interpret information on biodegradation of materials
- Infer evidence of an environmental change near a substance release
- Interpret LC50/toxicity data
- List domestic wastes that shouldn't be sent to regular landfills
- Evaluate methods for transporting, storing, and disposing of household chemicals
- Evaluate consumer and industrial practices for their impact on the environment

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