FINAL EXAM Summary Sheet

Potential Study Questions

<u>Disclaimer</u>: The following are potential topics for exam questions and sample questions; the exam may or may not contain these questions, and questions not included in this sheet may or may not be on the exam. There also may be typos in this sheet.

ALKANES

Physical Properties

- Hydrocarbons, dispersion forces, non-polar, not miscible w/water, low density
- sp³ hybridized carbon: C-C, C-H bonds, electronegativity, MO theory & orbitals
- m.p.: odd/even effect.
- b.p. C1-4: gases, C5-17: liquids, C18 and up: solids
- linear, branched, cyclics, chirality (incl. optical activity, D/L, R/S, d/l,+/-),
- non-equivalent carbons
- isomers: constitutional, conformational isomers, stereo-isomers
- naming, special groups:
 - o Me, Et, Pr, Bu
 - o iPr, iBu, sec-Bu, tBu
 - o neopentyl

<u>Task</u>: Name compounds 1 and 2; identify any chiral centers and # of chemically non-equivalent carbons; where would the compounds absorb in IR in UVvis? Identify the participating orbitals.



- Rotational/conformational isomers = conformations
 - o eclipsed
 - values: H/H, CH3/H, CH3/CH3
 - o staggered, anti, gauche
 - value: gauche: CH3/CH3

Task: draw an energy rotational energy plot (360°) for pentane along C2-C3 bond

- Cycloalkanes:
 - o Spiro, fused, bridged: naming
 - o Heats of combustion: ring strains (angle, torsional, sterics)
 - o Cis/trans
 - 1,3-diaxial interactions
 - value H/CH3
 - Ring-flip for cyclohexane

Acs Organic Chemistry Exam Study Guide

Shawn R. Simonson

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literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

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