- According to the Bronsted—Lowry definition, which chemical species can function both as an acid and as a base?
  - a. CI = 6. SO<sub>2</sub> =
  - SO/\*\*
    NH/\*
  - d. HCO<sub>2</sub>
  - e. H<sub>2</sub>O\*
- 2. In the reaction CN<sup>-</sup> + H<sub>2</sub>O ← → HCN + OH<sup>-</sup> which is an acid—base conjugate pair?
  - a. H<sub>2</sub>O and HCN
  - b. H<sub>2</sub>O and OH<sup>-</sup>
  - c. ON and H<sub>2</sub>O
  - d. HCN and OH"
- Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX
  - a. It is less basic than a I M solution of NaY.
  - b. It is more basic than a I M solution of NaY.
  - It yields a neutral solution.
  - It is more concentrated than a 1 M solution of NaY.
- 4. The pH of a 0.03 M HCI solution is
  - a. 1.5
  - 6. 25
  - c. 3.5
  - d. 12.5
- The pH of a 1.0 x 10<sup>-3</sup> M Ba(OH)<sub>2</sub> solution at 25°C is
  - a. 2.7
  - b. 3.0
  - c. 11.0
  - d. 11.3
- 6. Which is the strongest acid?
  - a. HCIO
  - A. HCIO<sub>2</sub>
  - c. HOIO<sub>3</sub>
  - d. HClO<sub>4</sub>
- 7. Which salt reacts with water (hydrolyzes) to produce a basic solution?
  - a. NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>
  - NaNO<sub>3</sub>
  - c. NH<sub>f</sub>CI
  - d. BaSO<sub>4</sub>

8. The weakest of the bases listed is

Acid	Conjugate Base	Ka (Ionization Constant of Acid)				
HCL	CIT	100% ionized				
HSO <sub>4</sub>	SO <sub>4</sub> 2-	1.2 x 10°				
H <sub>2</sub> S HS <sup>-</sup> HS <sup>-</sup> S <sup>r-</sup>		5.7x10 *				
		1.2x10 <sup>-0</sup>				

- a. CIT
- # 1.00 T
- e. HS<sup>-</sup> d. S<sup>3-</sup>
- e. 50e<sup>2-</sup>
- 9. The oxide of which element will react with water to form the strongest acid?

Period	Main Groups							
	1	П	П	IV	V	VI	VII	0
First								
Second	X	Y	Z	P		Q	S	U
Third	WAY					R	T	0.40

- e. W
- 6. M
- c. P
- d. R.
- Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate KC<sub>2</sub>H<sub>2</sub>O<sub>2</sub>, is less alkaline than a 0.10 M solution of potassium evanide. KCN?
  - a. Hydrocyanic acid is a weaker acid than acetic acid.
  - Hydrocyanic acid is less soluble in water than acetic acid.
  - c. Cyanides are less soluble than acetaces
  - d. Acetic acid is a weaker acid than hydrocyanic acid
- In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH, the properties of the solution at the equivalence point will correspond exactly to the properties of
  - a. a 0.100 M sodium solution.
  - a 0.0500 M sodium hydroxide solution.
  - c. a 0.0500 M benzoic acid solution.
  - a 0.0500 M sodium benzoate solution.

# **Acs First Term General Chemistry Practice Exam**

**Ensheng Dong** 

## **Acs First Term General Chemistry Practice Exam:**

Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS, 2018-10-15 This guide is separated into first term and second term general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems Diversity, Equity, and Inclusion for Mathematics and Science Education: Cases and Perspectives Lin, Cheng-Yao, Sun, Li, 2025-06-04 Diversity equity and inclusion DEI are critical pillars for transforming mathematics and science education As classrooms diversify the need to address systemic barriers and create inclusive learning environments becomes more urgent Cases on DEI in STEM education highlight the real world challenges and strategies educators face in promoting equitable access to learning opportunities dismantling biases and empowering students from historically marginalized communities Further exploration may reveal powerful teaching tools and catalyze reflective practice and institutional change encouraging educators to critically examine their roles in shaping a more inclusive future in math and science Cases on Diversity Equity and Inclusion for Mathematics and Science Education explores key issues and concepts related to diversity equity and inclusion in mathematics and science classrooms It offers solutions and successful strategies for teaching and learning in mathematics and science This book covers topics such as inclusive classrooms K 12 education pre service teaching and is a useful resource for educators sociologists academicians researchers and scientists Visualization in Science Education John K. Gilbert, 2006-03-30 Visualization in Science Education draws on the insights from cognitive psychology science and education by experts from Australia Israel Slovenia UK and USA It unites these with the practice of science education particularly the ever increasing use of computer managed modelling packages especially in chemistry The first section explores the significance and intellectual standing of visualization The second section shows how the skills of visualization have been developed practically in science education This is followed by accounts of how the educational value of visualization has been integrated into university courses in physics genomics and geology The fourth section documents experimental work on the classroom assessment of visualization An endpiece summarises some of the research and development needed if the contribution of this set of universal skills is to be fully exploited at all levels and in all science subjects Building Synergy for High-Impact Educational Initiatives Janine Graziano, Lauren Chism Schmidt, 2016-05-17 Published in partnership with the Washington Center for Improving the Quality of Undergraduate Education First year seminars and learning communities are two of the

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