


15. After a certain long period of time, a 100 gram sample of Carbon 14, C_{14} , experiences radioactive decay to become 20 grams of C_{14} . Since C_{14} has a half-life of 5730 years, which equation can be used to solve for t , the time in years it took for the radioactive decay ?

A. $100 = 20 \left(\frac{1}{2} \right)^{\frac{t}{5730}}$

 B. $20 = 100 (2)^{\frac{t}{5730}}$

C. $20 = 100 \left(\frac{1}{2} \right)^{\frac{t}{5730}}$

D. $20 = 100 \left(\frac{1}{2} \right)^{\frac{5730}{t}}$

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