

Section A

[15 marks]

Answer all questions in this section.

- 1 Which molecule has carbon atoms with sp hybrid orbitals?

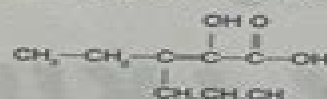
A $\text{CH}=\text{CH}$ **B** $\text{CH}_2=\text{CH}_2$ **C** CH_3CH_3 **D** C_6H_6

- 2 Combustion of hydrocarbon X in excess oxygen produces 0.66 g of carbon dioxide and 0.27 g of water. At room temperature and pressure, X is a gas with density 1.75 g dm^{-3} . What could the molecular formula of X be?

[The relative atomic mass: $\text{C} = 12$, $\text{H} = 1$; 1 mol of gas occupies 24 dm^3 at room temperature and pressure.]

A C_3H_4 **B** C_2H_6 **C** C_3H_8 **D** C_4H_8

- 3 What is the IUPAC name of the organic compound below?



A 2-Ethyl-1-hydroxy-1-hexenoic acid **C** 2-Hydroxy-3-propyl-2-pentenoic acid
B 3-Ethyl-2-hydroxy-2-hexenoic acid **D** 4-Hydroxy-3-propyl-3-pentenoic acid

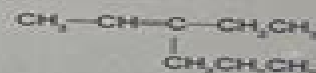
- 4 The following organic compounds show stereoisomerisms.



I



II



III

What type of stereoisomerism is shown by each compound?

I	II	III
A Geometrical	Geometrical	Optical
B Geometrical	Optical	Geometrical
C Optical	Geometrical	Optical
D Optical	Optical	Geometrical

- 5 The structural formulae of three organic acids are shown below.



I



II



III

Which is the correct sequence of the acids according to the ascending order based on their pK_a values?

A I, II, III **B** I, III, II **C** II, III, I

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