

## Acid, Base or Salt

Classify each of the following compounds as an acid, base or salt. Then, indicate whether each acid and base is strong or weak.

- |                              |       |       |
|------------------------------|-------|-------|
| 1. $\text{HNO}_3$            | _____ | _____ |
| 2. $\text{NaOH}$             | _____ | _____ |
| 3. $\text{NaNO}_3$           | _____ | _____ |
| 4. $\text{HCl}$              | _____ | _____ |
| 5. $\text{KCl}$              | _____ | _____ |
| 6. $\text{Ba(OH)}_2$         | _____ | _____ |
| 7. $\text{KOH}$              | _____ | _____ |
| 8. $\text{H}_2\text{S}$      | _____ | _____ |
| 9. $\text{Al(NO}_3)_3$       | _____ | _____ |
| 10. $\text{H}_2\text{SO}_4$  | _____ | _____ |
| 11. $\text{CaCl}_2$          | _____ | _____ |
| 12. $\text{H}_3\text{PO}_4$  | _____ | _____ |
| 13. $\text{Na}_2\text{SO}_4$ | _____ | _____ |
| 14. $\text{Mg(OH)}_2$        | _____ | _____ |
| 15. $\text{H}_2\text{CO}_3$  | _____ | _____ |
| 16. $\text{NH}_4\text{OH}$   | _____ | _____ |
| 17. $\text{NH}_4\text{Cl}$   | _____ | _____ |
| 18. $\text{HBr}$             | _____ | _____ |
| 19. $\text{FeBr}_3$          | _____ | _____ |
| 20. $\text{HF}$              | _____ | _____ |

# Acids Bases Salts Answers Pearson Education

**Roman Wölfel**



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