

## ACID & BASE

- Provide optimal cell function
  - Cells secrete **acid** and **bases** to maintain proper pH for enzymes to work.
- They both provide homeostatic regulation of pH of the body's ECF (extracellular fluid)
  - The Carbonic Acid-Bicarb Buffer System = pH homeostasis of blood

### ACID

CO<sub>2</sub> is regulated by the lungs + contributes to acid

- found in the stomach + urine + kidneys \*
  - ex: acid reflux, BPH, kidney failure

### BASE

Bicarb HCO<sub>3</sub> = form of carbon dioxide in our blood

- found in intestines \*
  - Regulated, excreted + reabsorbed in kidneys \*
  - Ex C-diff → bacteria in colon

## 2 FUNCTIONS

Metabolic + Respiratory

### METABOLIC

primary change is the **concentration of bicarbonate.**

#### METABOLIC **ACIDOSIS**

bicarb ↓ r/t the amount of acid in the body

- Increased Hydrogen ions
  - CAUSES: kidney failure, DKA, lactic acidosis (shock, sepsis), severe diarrhea
  - put pt. on bicarb drip to increase levels
- Respiratory compensation
  - Increased RR to rid of acid

#### METABOLIC **ALKALOSIS**

excess bicarb r/t the amount of acid in the body

- Decreased Hydrogen ions
  - CAUSES: loss of gastric fluids, antacid overdose
  - ex: too much NGT suctioning, hypokalemia, vomiting

### RESPIRATORY

primary change is the **concentration of carbonic acid**

#### RESPIRATORY **ACIDOSIS**

CO<sub>2</sub> is retained + inadequate O<sub>2</sub> exchange → increases amount of carbonic acid in the body

- CAUSES: Hypoventilation
- ex: Asthma, COPD, Pneumonia
- Decreased RR
  - Sleep apnea, drug overdose, head injuries

Compensation?

- Increasing RR

#### RESPIRATORY **ALKALOSIS**

Too much CO<sub>2</sub> is lost + carbonic acid levels fall

- Pt has increased RR
  - ex: hyperventilation due to anxiety attack

Compensation?

- slow + shallow breaths

# Acids Base Study Guide Content Mastery

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