

## Definitions

1 1. solution2 2. solute3 3. solvent4 4. solubility5 5. supersaturated solution6 6. electrolyte

a. capable of being dissolved

b. solution with excess in the solvent

c. substance that is dissolved in a solution

d. substance that dissolved in water for forms a solution that conducts an electric current

e. homogeneous mixture of two or more substances in a single physical state

f. substance that, upon dissolving, is a solution

## True/False Review

7. List 3 ways you can increase the rate of dissolving.

1. stir

2. heat

3. crush

8. Explain what the phrase "like dissolves like" means in relation to solution formation.

polar solvents + polar

nonpolar + polar or nonpolar

9. Explain how temperature affects the solubility of each of the following

a. solids  $\uparrow$  temp,  $\uparrow$  solubilityb. liquids  $\uparrow$  temp,  $\uparrow$  solubilityc. gases  $\uparrow$  temp,  $\downarrow$  solubility

10. Explain how pressure affects the solubility of each of the following

a. solids: doesn't

b. liquids: doesn't

c. gases  $\uparrow$  pressure,  $\uparrow$  solubility

# Acids And Bases Test Review Answers

**Barron's Educational Series, Neil D.  
Jespersen, Pamela Kerrigan**



## Acids And Bases Test Review Answers:

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