

Name: _____ Date: _____ Block: _____

Practice Acid/Base Quiz

Directions: Answer all questions on this practice quiz. Use this as a resource to study for the assessment on Friday, February 28th in addition to your lecture note guides and unit study guide. Your assessment will include both multiple choice and short answer questions.

1. Which of the following could have a pH of 10?

- a. HCl b. HCH_2COO c. NH_4OH d. HF

2. Which of the following is a neutralization reaction?

- a. $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$ c. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
b. $\text{Ca}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{Ca}(\text{NO}_3)_2$ d. $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$

3. Which of the following is directly responsible for acid rain?

- a. steam vented from a nuclear power plant
b. sulfur dioxide released from a coal-fired power plant
c. mining of coal for a coal-fired power plant
d. processing of uranium for a nuclear power plant

Use the following reaction to answer questions 4 through 8.



4. Which compound represents the acid?

- a. LiOH b. H_2O c. LiF d. HF

5. Which compound represents the salt?

- a. LiOH b. H_2O c. LiF d. HF

6. Which compound represents the base?

- a. LiOH b. H_2O c. LiF d. HF

7. Which compound has the highest number of hydrogen (H^+) ions?

- a. LiOH b. H_2O c. LiF d. HF

8. Which compound has the highest number of hydroxide (OH^-) ions?

- a. LiOH b. H_2O c. LiF d. HF

Acids And Bases Test Review Answers

**Barron's Educational Series, Neil D.
Jespersen, Pamela Kerrigan**



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