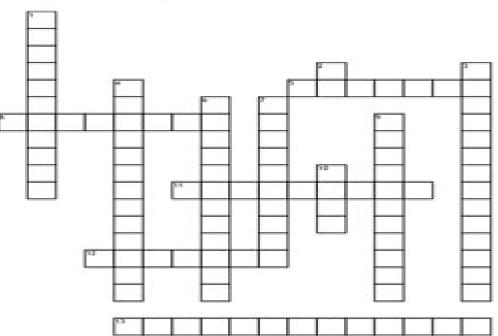
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# Acids, Bases, and Salts

#### Across

a logarithmic scale
that measures the acidity
or alkalinity of a solution
on a scale of 1 to 14
 the concentration of
a substance in moles per
liter of solution
 a substance that
undergoes a color change
that can be used to
determine when a reaction
is complete

12. a measure of the hydrogen (hydronium) ion concentration of a solution 13. a substance that produces hydroxide ions (OH-) as the only negative ions when dissolved in water



#### Down 1 a si

1. a substance whose water solution conducts an electric current

a logarithm of a solution's hydrogen ion concentration

the reaction between an acid and a base to produce water and salt

 a substance that produces hydronium ions (H3O+) as the only positive ions when dissolved in water  A compound formed by the combination of water with a hydrogen ion, H3O+

7. a measure of the hydroxide ion concentration of a solution

 a hydrogen atom without it's electron (consisting solely on a proton)

10. the product other than water of a neutralization reaction; an ionic substance consisting of a metallic cation and anion other than the hydroxide ion



# **Acids And Bases Crossword Answers**

**AW Rasmussen** 

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