

Acidity of Beverages Lab

Introduction: Common beverages may be either acidic or basic. Fruit juices, for example, get their sweet taste from sugars and their sour or tart taste from weak acids such as citric acid. If the juice contains too much sugar, it will taste bland, but too much acid and the juice will taste sour. The concentration of acids in various consumer beverages may be determined by titration with sodium hydroxide.

Background: The main acids present in fruits and fruit juices are citric acid (in citrus fruits), tartaric acid (in grapes), and malic acid (in apples). All of these are characterized as weak acids.

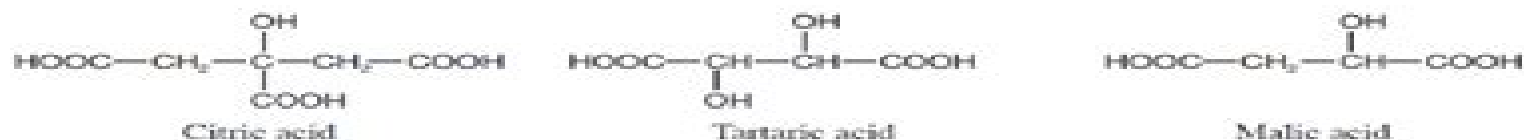
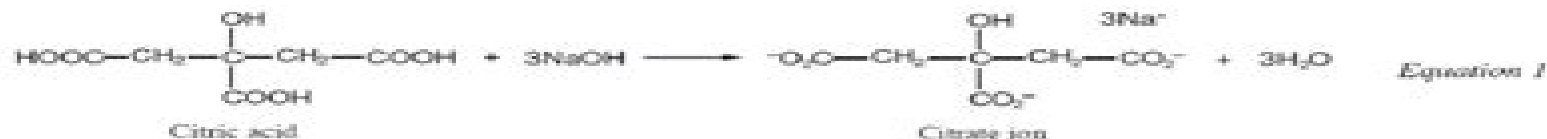


Figure 1. Organic acids in fruits and fruit juices

The amount of citric acid in citrus fruit juices can be determined by titration with a standard solution of sodium hydroxide. A standard solution is one whose concentration is accurately known, usually to three significant figures. Citric acid is a tricarboxylic acid—it has three ionizable or “active” hydrogen atoms in its structure. One mole of citric acid therefore reacts with three moles of sodium hydroxide via the acid-base neutralization reaction shown in Equation 1.



Acid-base titrations are an extremely useful technique to determine the concentration of an acid or base in a sample. In titrating beverages such as orange juice, apple juice, and sodas that contain weak acids, the juice is called the analyte and a strong base is used as the titrant.

In the titration procedure, a sodium hydroxide solution of known molarity is carefully added using a buret to a measured volume of fruit juice containing an indicator. The exact volume of sodium hydroxide that must be added to reach the indicator endpoint is measured and then used to calculate the concentration of citric acid in the juice.

A sample setup for a titration is shown in Figure 2 on the next page, where a buret containing the titrant is clamped to the support stand and a beaker or flask containing the analyte is set on a stir plate. If a pH probe is inserted into the solution, a titration curve can be constructed by plotting the pH of the solution on the y-axis versus the volume of titrant added on the x-axis. The shape of the titration curve may be used to distinguish strong and weak acids in the analyte, and also permits graphical analysis of the equivalence point. At the equivalence point, moles of added titrant are stoichiometrically related to moles of analyte in the sample.

Acidity Of Beverages Lab Anser

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Handbook of Alcoholic Beverages, 2 Volume Set Alan J. Buglass, 2011-02-14 HANDBOOK OF ALCOHOLIC BEVERAGES A comprehensive two volume set that describes the science and technology involved in the production and analysis of alcoholic beverages HANDBOOK OF ALCOHOLIC BEVERAGES Technical Analytical and Nutritional Aspects At the heart of all alcoholic beverages is the process of fermentation particularly alcoholic fermentation whereby sugars are converted to ethanol and many other minor products The Handbook of Alcoholic Beverages tracks the major fermentation process and the major chemical physical and technical processes that accompany the production of the world s most familiar alcoholic drinks Indigenous beverages and small scale production are also covered to a significant extent The overall approach is multidisciplinary reflecting the true nature of the subject Thus aspects of biochemistry biology including microbiology chemistry health science nutrition physics and technology are all necessarily involved but the emphasis is on chemistry in many areas of the book Emphasis is also on more recent developments and innovations but there is sufficient background for less experienced readers The approach is unified in that although different beverages are dealt with in different chapters there is extensive cross referencing and comparison between the subjects of each chapter Appropriate for food professionals working in the development and manufacture of alcohol based drinks as well as academic and industrial researchers involved in the development of testing methods for the analysis and regulation of alcohol in the drinks industry Divided into five parts this comprehensive two volume work presents INTRODUCTION BACKGROUND AND HISTORY a simple introduction to the history and development of alcohol and some recent trends and developments FERMENTED BEVERAGES BEERS CIDERS WINES AND RELATED DRINKS the latest innovations and aspects of the different fermentation processes used in beer wine cider liqueur wines fruit wines low alcohol and related beverages SPIRITS covers distillation methods and stills used in the production of whisky cereal and cane based spirits brandy fruit spirits and liqueurs ANALYTICAL METHODS covering the

monitoring of processes in the production of alcoholic beverages as well as sample preparation chromatographic spectroscopic electrochemical physical sensory and organoleptic methods of analysis NUTRITION AND HEALTH ASPECTS RELATING TO ALCOHOLIC BEVERAGES includes a discussion on nutritional aspects both macro and micro nutrients of alcoholic beverages their ingestion absorption and catabolism the health consequences of alcohol and details of the additives and residues within the various beverages and their raw materials Illustrated Guide to Home Chemistry Experiments

Robert Bruce Thompson,2008-04-29 Provides information on setting up an in home chemistry lab covers the basics of chemistry and offers a variety of experiments **Beverage World** ,2002 **Educart CBSE Class 12 BIOLOGY One Shot Question Bank 2024-25 (Updated for 2025 Exam)** Educart,2024-06-28 *Forensics and Applied Science Experiments* ,2005 **Educart CBSE Class 12 Biology One Shot Question Bank 2026 (Includes PYQs for 2025-26)**

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Janardhanan.T,Sanjay Sharma,2021-07-26

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