

Assessment

Acids and Bases

Section Quiz: Acid-Base Theories

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. All Brønsted- Lowry acids
 - a. are aqueous solutions.
 - b. can act as Arrhenius acids.
 - c. donate protons.
 - d. All of the above

- _____ 2. Which of the following substances is both a Brønsted-Lowry base and an Arrhenius base?
 - a. $\text{NH}_3(s)$
 - b. $\text{NH}_3(aq)$
 - c. $\text{HCl}(g)$
 - d. $\text{HCl}(aq)$

- _____ 3. In the following reaction, which substance acts as a Brønsted-Lowry acid?

$$\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightarrow \text{HSO}_4^- + \text{H}_3\text{O}^+$$
 - a. H_2SO_4
 - b. H_2O
 - c. HSO_4^-
 - d. H_3O^+

- _____ 4. In the following reaction, which substance acts as a Brønsted-Lowry base?

$$\text{HCl} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Cl}^-$$
 - a. HCl
 - b. H_2O
 - c. H_3O^+
 - d. Cl^-

- _____ 5. Which of the following is a conjugate acid?
 - a. HCl
 - b. H_2O
 - c. H_3O^+
 - d. OH^-

Acid Base Theories Section Review Answers

Mary Lou Turgeon



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