#### CHAPTER 26. : ACID-BASE

Total body water volume: \_\_\_\_\_this is 60% of body weight

Water: universal solvent

Solutes: nonelectrolytes and electrolytes

#### Extracellular and intracellular

- ECF : the fluid on the \_\_\_\_\_\_ the cell major cation is sodium and major anion is chloride
- Sodium is the most abundant cation in the extracellular fluid
- ICF: fluid \_\_\_\_\_ the cell major cation is potassium

Water intake is \_\_\_\_\_\_to water output 2500 ml/day
Water output urine, insensible water loss, perspiration, and feces
Water reabsorption: in collecting ducts proportional to ADH release

Aldosterone: important in regulation of sodium ion concentration "water follows sodium

Decrease ADH: diluted urine and decrease in volume in body fluids Increase ADH: concentrated urine

Atrial natriuretic peptide: decrease BP and blood volume

#### Important buffer systems

- Respiratory system
- Kildneys

#### normal pH of body fluids

- Arterial blood = pH 7.4
- Venous blood and IF fluid =pH 7.35
- ICF = pH 7.0
- Alkalosis or alkalemia = arterial blood pH > 7.45
- Acidosis or acidemia = arterial pH < 7.35</li>
- Physiological acidosis = pH between 7.0 7.35

#### Hydrogen ion regulation

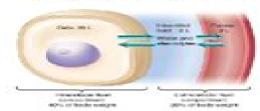
- Chemical buffer system \_\_\_\_\_\_ (rapid) first line of defense
- 2. Respiratory centers in brain stem act within 1 3 min
- 3. Renal mechanisms \_\_\_\_\_most potent

Three major chemical buffers

Chemical buffer, resist change in pH changes when a strong acid or base is added

- Bicarbonate buffer system.
- Phosphate buffer system
- 3. Protein buffer system

#### Major fluid compartments in the body





# **Acid And Base Study Guide Answers**

**Katheryn Evans** 

#### **Acid And Base Study Guide Answers:**

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