

AP Chemistry Summer Assignment

Summer 2014

All documents will be linked to the class web-site. As well, the answers will be linked for the practice test, but not the solutions. Solutions by you the students will be required to receive credit for the work.

Part 1

- Sketch all the **lab equipment** listed below and identify each piece's purpose. There are websites that contain this information, you need to search for them.
- | | | |
|---|-----------------------|--------------------|
| 1. Erlenmeyer Flask | 12. Filter Paper | 24. Crucible Tongs |
| 2. Pasteur Pipette | 13. Evaporating Dish | 25. Thermometer |
| 3. Glass Stir Rod | 14. Watch Glass | 26. Wire Gauze |
| 4. Volumetric Flask | 15. Mortar and Pestle | 27. Wash Bottle |
| 5. Beakers | 16. Crucible and Lid | 28. Wire Brush |
| 6. Graduated Cylinder | 17. Clay Triangle | 29. Beaker Tongs |
| 7. Test Tubes | 18. Buret Clamp | 30. Spark Lighter |
| 8. Rubber Stoppers (with and without holes) | 19. Utility Clamp | 31. Metal Scoop |
| 9. Eudiometer | 20. Ring Stand | |
| 10. Buret | 21. Ring Clamp | |
| 11. Funnel | 22. Bunsen Burner | |
| | 23. Test Tube Clamp | |
- Memorize the names and formulas of the **common polyatomic ions**, on page 65 of the book (Table 2.5, Common Polyatomic Ions). I recommend making flashcards and/or use Quizlet. Please keep your flashcards, as you will be tested on them throughout the entire year.
 - There will be a test on the FIRST DAY OF SCHOOL on the polyatomic ions!

Part 2

Complete the practice test by the first day of class. There will be a test on this material within the first two weeks of school

Summer Review Practice Test

- Answer on a separate sheet of paper. Show all work (for problems that require it)
A. Formula
B. Work with units
C. Answer with correct units
- Determine the number of significant figures in the following numbers.
A. 0.02
B. 0.020
C. 501
D. 501.0
E. 5000
F. 5000.
- In your laboratory experiment, you determine that the molar mass of a compound is 34.45 grams/mol. The actual molar mass of this compound is 43.25 grams/mol. What is your percent error?
- How many protons, neutrons, and electrons do the following atoms or ions have?
A. Ca
B. I⁻
C. uranium
D. Fe³⁺
E. carbon-13
- Identify the type of bond in the following compounds (ionic, covalent, metallic):
A. CO₂
B. MgCl₂
C. Fe-Fe
D. FeO
- Name the following ionic compounds:
A. NaCl
B. MgBr₂
C. FeO
D. Co(NO₃)₂

13 Ap Chemistry Summer Assignment Answers

Huangqi Zhang



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Table of Contents 13 Ap Chemistry Summer Assignment Answers

1. Understanding the eBook 13 Ap Chemistry Summer Assignment Answers
 - The Rise of Digital Reading 13 Ap Chemistry Summer Assignment Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 13 Ap Chemistry Summer Assignment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 13 Ap Chemistry Summer Assignment Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 13 Ap Chemistry Summer Assignment Answers
 - Personalized Recommendations
 - 13 Ap Chemistry Summer Assignment Answers User Reviews and Ratings
 - 13 Ap Chemistry Summer Assignment Answers and Bestseller Lists
5. Accessing 13 Ap Chemistry Summer Assignment Answers Free and Paid eBooks

- 13 Ap Chemistry Summer Assignment Answers Public Domain eBooks
- 13 Ap Chemistry Summer Assignment Answers eBook Subscription Services
- 13 Ap Chemistry Summer Assignment Answers Budget-Friendly Options
- 6. Navigating 13 Ap Chemistry Summer Assignment Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 13 Ap Chemistry Summer Assignment Answers Compatibility with Devices
 - 13 Ap Chemistry Summer Assignment Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 13 Ap Chemistry Summer Assignment Answers
 - Highlighting and Note-Taking 13 Ap Chemistry Summer Assignment Answers
 - Interactive Elements 13 Ap Chemistry Summer Assignment Answers
- 8. Staying Engaged with 13 Ap Chemistry Summer Assignment Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 13 Ap Chemistry Summer Assignment Answers
- 9. Balancing eBooks and Physical Books 13 Ap Chemistry Summer Assignment Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 13 Ap Chemistry Summer Assignment Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 13 Ap Chemistry Summer Assignment Answers
 - Setting Reading Goals 13 Ap Chemistry Summer Assignment Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 13 Ap Chemistry Summer Assignment Answers
 - Fact-Checking eBook Content of 13 Ap Chemistry Summer Assignment Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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