

7. Alkanols can be manufactured from alkenes by the initial reaction of alkenes with
- bromine in tetrachloromethane.
 - concentrated tetraoxosulphate(VI) acid.
 - aqueous potassium tetraoxomanganate(VII).
 - sodium hydroxide solution.
8. Which of the following statements about the standard hydrogen electrode is **not** correct?
- The hydrogen gas is at a pressure of 1 atmosphere
 - A solution containing 1 mol dm⁻³ of H⁺_(aq) ions is used
 - A platinum electrode is used
 - The temperature is kept at 20 °C
9. If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be? [O = 16.0, M = 120]
- MO
 - MO₂
 - MO₃
 - MO₄
10. The products of the electrolysis of dilute sodium chloride using carbon electrodes are
- | | <u>Anode</u> | <u>Cathode</u> |
|----|--------------|----------------|
| A. | chlorine | sodium. |
| B. | oxygen | hydrogen. |
| C. | chlorine | hydrogen. |
| D. | hydrogen | oxygen. |
11. Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.
- 12 C
 - 120 C
 - 360 C
 - 720 C
12. Oxochlorate(I) acid is used as a bleaching agent because it is
- a weak acid.
 - a reducing agent.
 - an oxidizing agent.
 - a strong acid.
13. The IUPAC name for CH₃CHCH₃CHClCHCH₃CH₂CH₃ is
- 2, 4 - dimethyl - 3 - chlorohexane.
 - 3, 5 - dimethyl - 4 - chlorohexane.
 - 4 - chloro - 3, 5 - dimethylhexane.
 - 3 - chloro - 2, 4 - dimethylhexane.
14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn
- acidified K₂Cr₂O₇ solution colourless.
 - acidified KMnO₄ solution colourless.
 - Fe(NO₃)₂ solution green.
 - Pb(NO₃)₂ paper black.
15. If 5.0 g of marble reacts with 25.0 cm³ hydrochloric acid, which of the following combinations has the fastest reaction rate?
- Marble chips and 2.0 mol dm⁻³ HCl_(aq)
 - Marble chips and 2.5 mol dm⁻³ HCl_(aq)
 - Powdered marble and 2.5 mol dm⁻³ HCl_(aq)
 - Powdered marble and 2.0 mol dm⁻³ HCl_(aq)

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 - The Rise of Digital Reading Waec Question And Answer In Chemistry Main Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Waec Question And Answer In Chemistry Main Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Waec Question And Answer In Chemistry Main Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Waec Question And Answer In Chemistry Main Paper
 - Personalized Recommendations
 - Waec Question And Answer In Chemistry Main Paper User Reviews and Ratings

- Waec Question And Answer In Chemistry Main Paper and Bestseller Lists
- 5. Accessing Waec Question And Answer In Chemistry Main Paper Free and Paid eBooks
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 - Waec Question And Answer In Chemistry Main Paper Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
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 - Waec Question And Answer In Chemistry Main Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Waec Question And Answer In Chemistry Main Paper
 - Highlighting and Note-Taking Waec Question And Answer In Chemistry Main Paper
 - Interactive Elements Waec Question And Answer In Chemistry Main Paper
- 8. Staying Engaged with Waec Question And Answer In Chemistry Main Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Waec Question And Answer In Chemistry Main Paper
- 9. Balancing eBooks and Physical Books Waec Question And Answer In Chemistry Main Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Waec Question And Answer In Chemistry Main Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Waec Question And Answer In Chemistry Main Paper
 - Setting Reading Goals Waec Question And Answer In Chemistry Main Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Waec Question And Answer In Chemistry Main Paper
 - Fact-Checking eBook Content of Waec Question And Answer In Chemistry Main Paper
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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