

Name \_\_\_\_\_

*Chemistry Worksheet*  
*Naming & Formula Writing (Ionic)*

**Instructions:** Write the formulas &/or the names for the compounds listed below

1. Sodium nitrate \_\_\_\_\_
2. Calcium carbonate \_\_\_\_\_
3. Magnesium oxide \_\_\_\_\_
4. Ammonium sulfide \_\_\_\_\_
5. Lead (II) sulfate \_\_\_\_\_
6. Sodium cyanide \_\_\_\_\_
7. Potassium hydroxide \_\_\_\_\_
8. Silver chloride \_\_\_\_\_
9. Iron (III) hydroxide \_\_\_\_\_
10. Potassium hydroxide \_\_\_\_\_
11. Tin (IV) perchlorate \_\_\_\_\_
12. Potassium carbonate \_\_\_\_\_
13. Silver nitrate \_\_\_\_\_
14. Sodium iodide \_\_\_\_\_
15. Ammonium hydroxide \_\_\_\_\_
16. Potassium iodate \_\_\_\_\_
17. Lead (IV) oxide \_\_\_\_\_
18. Ammonium hydroxide \_\_\_\_\_
19. Barium sulfate \_\_\_\_\_
20. barium chloride \_\_\_\_\_
21. Cobalt (II) chloride \_\_\_\_\_
22. Sodium carbonate \_\_\_\_\_
23. Calcium oxide \_\_\_\_\_
24. Lead (II) nitrate \_\_\_\_\_
25. Tin (II) chloride \_\_\_\_\_

26. Aluminum chloride \_\_\_\_\_
27. Iron (III) hydroxide \_\_\_\_\_
28. Sodium acetate \_\_\_\_\_
29. calcium hydroxide \_\_\_\_\_
30. sodium iodate \_\_\_\_\_
31. Nickel (II) nitrate \_\_\_\_\_
32. Iron (II) chloride \_\_\_\_\_
33. Magnesium bromide \_\_\_\_\_
34. Ammonium nitrate \_\_\_\_\_
35. Silver bromide \_\_\_\_\_
36.  $Al(OH)_3$  \_\_\_\_\_
37.  $NH_4I$  \_\_\_\_\_
38.  $Li_2CO_3$  \_\_\_\_\_
39.  $CuSO_4$  \_\_\_\_\_
40. KCN \_\_\_\_\_
41.  $Pb(ClO)_2$  \_\_\_\_\_
42. BaS \_\_\_\_\_
43.  $ZnSO_4$  \_\_\_\_\_
44.  $Pb(CH_3COO)_2$  \_\_\_\_\_
45.  $Ca(NO_3)_2$  \_\_\_\_\_
46.  $Fe_2(CO_3)_3$  \_\_\_\_\_
47.  $NH_4IO_3$  \_\_\_\_\_
48.  $CaCl_2$  \_\_\_\_\_
49. NaF \_\_\_\_\_
50.  $Cu(NO_3)_2$  \_\_\_\_\_

# 33 Physical Science Names And Formulas Worksheet

**James T. Shipman**



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## **Table of Contents 33 Physical Science Names And Formulas Worksheet**

1. Understanding the eBook 33 Physical Science Names And Formulas Worksheet
  - The Rise of Digital Reading 33 Physical Science Names And Formulas Worksheet
  - Advantages of eBooks Over Traditional Books
2. Identifying 33 Physical Science Names And Formulas Worksheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 33 Physical Science Names And Formulas Worksheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from 33 Physical Science Names And Formulas Worksheet
  - Personalized Recommendations
  - 33 Physical Science Names And Formulas Worksheet User Reviews and Ratings
  - 33 Physical Science Names And Formulas Worksheet and Bestseller Lists
5. Accessing 33 Physical Science Names And Formulas Worksheet Free and Paid eBooks
  - 33 Physical Science Names And Formulas Worksheet Public Domain eBooks
  - 33 Physical Science Names And Formulas Worksheet eBook Subscription Services
  - 33 Physical Science Names And Formulas Worksheet Budget-Friendly Options

6. Navigating 33 Physical Science Names And Formulas Worksheet eBook Formats
  - ePub, PDF, MOBI, and More
  - 33 Physical Science Names And Formulas Worksheet Compatibility with Devices
  - 33 Physical Science Names And Formulas Worksheet Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 33 Physical Science Names And Formulas Worksheet
  - Highlighting and Note-Taking 33 Physical Science Names And Formulas Worksheet
  - Interactive Elements 33 Physical Science Names And Formulas Worksheet
8. Staying Engaged with 33 Physical Science Names And Formulas Worksheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 33 Physical Science Names And Formulas Worksheet
9. Balancing eBooks and Physical Books 33 Physical Science Names And Formulas Worksheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 33 Physical Science Names And Formulas Worksheet
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 33 Physical Science Names And Formulas Worksheet
  - Setting Reading Goals 33 Physical Science Names And Formulas Worksheet
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 33 Physical Science Names And Formulas Worksheet
  - Fact-Checking eBook Content of 33 Physical Science Names And Formulas Worksheet
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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