

7. One gram of steam at 100°C condenses, and the water cools to 22°C .

a. How much heat is released when the steam condenses? 540

b. How much heat is released when the water cools from 100°C to 22°C ?

78 cal ($100 - 22 \times 1 \text{ cal/g}^{\circ}\text{C}$)

c. How much heat is released altogether?

618 cal



8. In a household radiator 1000 g of steam at 100°C condenses, and the water cools to 90°C .

a. How much heat is released when the steam condenses?

540,000 cal

b. How much heat is released when the water cools from 100°C to 90°C ?

10,000 cal

c. How much heat is released altogether?

550,000 cal



9. Radioactive minerals in common granite release 0.01 cal/kg of energy per year. If a 1-kg chunk of 50°C granite is thermally insulated, so all this energy heats it, how many years does it take to reach its melting temperature of 700°C ? (Assume the specific heat of granite is $200 \text{ cal/kg}^{\circ}\text{C}$.)

$Q = mc\Delta T = 1 \text{ kg} (200 \text{ cal/kg}^{\circ}\text{C}) (650^{\circ}\text{C}) = 130,000 \text{ cal} / 0.10 \text{ cal/year} = 1,300,000 \text{ yr}$

10. How many years would be required if the chunk of granite had a mass of 1 million kg? Why?

SAME! (MORE ROCK = MORE RADIATION)

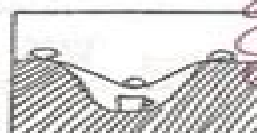
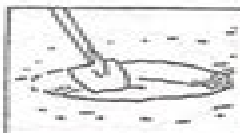
11. To calculate the time it takes to melt the 700°C granite, what other information would you need?

HEAT OF FUSION FOR GRANITE

12. So we see that radioactivity keeps the earth's interior hot. After energy due to radioactivity eventually migrates to the earth's surface, in what form does it leave the earth?

TERRESTRIAL RADIATION

13. To get water from the ground, even in the hot desert, dig a hole about a half meter wide and a half meter deep. Place a cup at the bottom. Spread a sheet of plastic wrap over the hole and place stones along the edge to hold it secure. Weight the center of the plastic with a stone so it forms a cone shape. Why will water collect in the cup? (Physics can save your life if you're ever stranded in a desert!)



① Terr. Radiation makes water evapor.
② Evap. water trapped on plastic
③ Condensed water runs into cup.

(Some also will collect at night on the top of the sheet)

Print!

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Paul W Zitzewitz



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Todd,2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between **Advanced**

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