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web equilibrium moles hscn 2.00×10^{-5} mol 3.00×10^{-6} mol 1.70×10^{-5} mol hscn knowing the number of moles of Fe^{3+} and hscn present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species mol Fe^{3+} 1.70×10^{-5} mol Fe^{3+} 8.50×10^{-4} M

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explain the meaning of the statement equilibrium is macroscopically static but microscopically dynamic very

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