4 (Chemical Equations and Stoichiometry

SIX STOICHIOMETRY PROBLEMS

General Stoichiometry

[Molar masses:

[Molar masses:

Several brands of antacid tablets use aluminum hydroxide to neutralize excess acid.

What quantity of HCl, in grams, can a tablet with 0.750 g of Al(OH)₃ consume? What quantity of water is produced?

1.05 g HCI

0.520 g H₂O

17. The equation for one of the reactions in the process of reducing iron ore to the metal is

(a) What is the maximum mass of iron, in grams, that can be obtained from 454 g (1.00 lb) of iron(III) oxide?

31.8 g Fe

(b) What mass of CO is required to reduce the iron(III) oxide to iron metal?

239 g CO

Limiting Reactants

The reaction of methane and water is one way to prepare hydrogen:

If you begin with 995 g of CH_s and 2510 g of water, what is the maximum possible yield of H₂?

CH4 is LR.... 376 g H2 is formed

[Molar masses:

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