

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.

10. A cation has a positive charge.

11. A binary ionic compound contains only one kind of cation and one kind of anion.

12. Monatomic ions consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is produced when an ionic compound is formed.

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R Bogdan



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Math Phonics - Multiplication (ENHANCED eBook) Marilyn B. Hein, 1996-03-01 Math Phonics is a specially designed program for teaching the mastery of basic math concepts and facts The name Math Phonics tm is used because the rules patterns and memory techniques developed for this program are similar to those used in language arts Most of the rules are short and easy to learn Children are taught to look for patterns and use them Repetition and drill are the keys In just minutes a day your students can master the multiplication facts 0 through 12 **Applied Behavior Analysis** Edward P.

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students and relates to their lives From decades of experience teaching the Hockenburys created a book filled with cutting edge science and real life stories that draw students of all kinds into the course **Motivated to Learn** Staci M.

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