

NAME: _____

CLASS: _____

DATE: _____

1. If a chemical reaction such as photosynthesis begins with 6 atoms of carbon (C), how many atoms of carbon (C) should be in the products?

- | | | | |
|-------------------------|----------------------|-------------------------|-----------------------|
| <input type="radio"/> A | b) 6 atoms of carbon | <input type="radio"/> B | a) 12 atoms of carbon |
| <input type="radio"/> C | d) 2 atoms of carbon | <input type="radio"/> D | c) 3 atoms of carbon |

2. How does a balanced chemical equation satisfy the Law of Conservation of Mass?

- | | | | |
|-------------------------|---|-------------------------|--|
| <input type="radio"/> A | a) During a chemical reaction, the total amount of matter stays the same. | <input type="radio"/> B | c) During a chemical reaction, one or more new substances is formed. |
| <input type="radio"/> C | b) During a chemical reaction, matter is destroyed. | <input type="radio"/> D | d) During a chemical reaction, the total number of atoms increases. |

3. When wood is burned, it is changed into ash and carbon is released into the atmosphere as smoke. If 2 kg of wood is burned, how much ash would result?

- | | | | |
|-------------------------|---|-------------------------|---|
| <input type="radio"/> A | b) Less than 2 kg because some of the mass is lost as gas. | <input type="radio"/> B | a) Exactly 2 kg because mass is conserved in this system. |
| <input type="radio"/> C | c) Greater than 2 kg because carbon is added to the system. | <input type="radio"/> D | d) About 2 kg because some carbon escapes the system and some ash is added. |

4. A compound containing potassium (K), chlorine (Cl) and oxygen (O) has the chemical formula KClO_3 . Which best represents a balanced chemical reaction showing the breakdown of KClO_3 into simpler substances?

- | | | | |
|-------------------------|---|-------------------------|---|
| <input type="radio"/> A | c) $2\text{KClO}_3 \rightarrow 2\text{K} + 3\text{O}_2$ | <input type="radio"/> B | b) $2\text{KClO}_3 \rightarrow \text{Cl}_2 + 3\text{O}_2$ |
| <input type="radio"/> C | a) $2\text{KClO}_3 \rightarrow 2\text{KCl}$ | <input type="radio"/> D | d) <u>$2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$</u> |

5. Which best compares a homogeneous mixture and a heterogeneous mixture?

- | | |
|------------------------------------|---|
| a) Homogeneous mixtures are formed | c) The particles in a homogeneous mixture will separate if left |
|------------------------------------|---|

6th Grade Chemistry Test Review

Oscar Krisen Buros



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