SC06_GROW_CR02 6/5/26 8:49 AM Page 17

Name Class Date

Section 2-2 Properties of Water (pages 40-43)

Key Concepts

- Why are water molecules polar?
- · What are acidic solutions? What are basic solutions?

The Water Molecule (pages 40-41)

- Is the following sentence true or false? A water molecule is neutral.
- Why is a water molecule polar? There is an uneven distribution of electrons between the oxygen and hydrogen atoms.
- Circle the letter of each sentence that is true about hydrogen bonds.
 - a. A hydrogen bond is stronger than an ionic bond.
 - (b) The attraction between the hydrogen atom on one water molecule and the oxygen atom on another water molecule is an example.
 - c. A hydrogen bond is stronger than a covalent bond.
 - (d.) They are the strongest bonds that form between molecules.
- 4. Complete the table about forms of attraction.

FORMS OF ATTRACTION

Form of Attraction	Definition
Cohesion	Attraction between molecules of the same substance
Adhesion	Attraction between molecules of different substances

Solutions and Suspensions (pages 41-42)

- What is a mixture? A mixture is a material composed of two or more elements or compounds that are physically mixed together but not chemically combined.
- 7. The greatest solvent in the world is _______
- What is a suspension? A suspension is a mixture of water and nondissolved substances that are so small they do not settle out.



2 2 Properties Of Water Answers

Camilla Rothe

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decrease depending upon organizational requirements The last candidate securing equal marks in Tier I in the bracket will also be included Merit and final selection will be made based on the performance of the candidates in Tier I Subjects Covered Quantitative Aptitude Numerical Ability General Awareness Reasoning Ability Exam Patterns Phase I CBT Screening The total duration is 90 minutes for stage Tier 1 The question paper will be available in the English and Hindi language For each correct question the candidate will be awarded a 1 mark Negative Marking NO Conducting Body Defence Research Development Organisation Centre for Personnel Talent Management Exam Exam Level National Exam Category and Exam Board Defence Research Development Organisation Centre for Personnel Talent Management Exam Science Tenth Class Part 2 Chemistry (Solutions of Lakhmir Singh & Manjit Kaur) Amar Bhutani, This book includes the answers to the questions given in the textbook CBSE Science Tenth Class Part 2 Chemistry published by S Chand Co and written by Lakhmir Singh and Manjit Kaur This book is based for latest syllabus **Objective Chemistry Vol 2 For** Engineering Entrances 2022, 1 Complete Study Pack for Engineering Entrances series provides Objective Study Guides 2 Objective Chemistry Volume 2 is prepared in accordance with NCERT Class 11th syllabus 3 Guide is divided into 25 chapter 4 complete text materials Practice Exercises and workbook exercises with each theory 5 Includes more than 5000 MCQs collection of Previous Years Solved Papers of JEE Main and Advanced BITSAT Kerala CEE KCET AP Objective Chemistry Volume 2 is divided into 25 chapters giving Complete Text Material along with Practice Exercises and Workbook exercises Chapter Theories are coupled with well illustrated examples helping students to learn the basics of Chemistry Housed with more than 5000 MCQs and brilliant collection of Previous Years Solved Papers of JEE Main and Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT and MHT CET which is the most defining part of this book Delivering the invaluable pool of study resources for different engineering exams at one place this is no doubt an excellent book to maximize your chances to get qualified at engineering entrances TOC Solid State Solutions Electrochemistry Chemical Kinetics Surface Chemistry Chemical Kinetics Surface Chemistry General Principle and Processes of Isolation of Elements p Block Elements I Group 15 p Block Elements II Group 16 p Block Elements III Group 17 p Block Elements IV Group 18 d and f block Elements Coordinate Compounds Haloalkanes Haloarenes Alcohols Phenols Ether Aldehyes and Ketones Carboxylic Acids Amines Diazonium Salts Cyanides and Isocyanides Bimolecules Polymers Chemistry in Everyday Life Principles Related to Practical Chemistry JEE Advanced Solved Paper 2015 JEE Main Advanced Solved Papers 2016 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2017 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2018 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2019 20

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