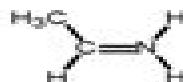


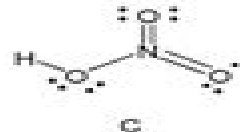
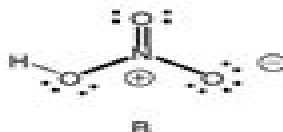
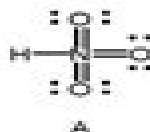
## PRACTICE PROBLEMS, CHAPTERS 1 - 3

(Covered from Ch. 3: Alkane and Alkyl Halide nomenclature only)

- The atomic number of boron is 5. The correct electronic configuration of boron is:  
 A.  $1s^2 2s^3$       B.  $1s^2 2p^3$       C.  $1s^2 2s^2 2p^1$       D.  $2s^2 2p^3$       E.  $1s^2 2s^2 3s^1$
- How many distinct  $p$  orbitals exist in the second electron shell, where  $n = 2$ ?
- An oxygen atom has \_\_\_\_\_ valence electrons.
- Draw a correct Lewis structure for boric acid,  $B(OH)_3$ .
- Draw a correct Lewis structure for *t*-butanol,  $(CH_3)_3COH$ .
- Draw a correct Lewis structure for acetonitrile,  $CH_3CN$ .
- The formal charge on nitrogen in the compound below is \_\_\_\_\_.



- Which of the following are correct Lewis structures for nitric acid,  $HNO_3$ ?



- The formal charge on oxygen in dimethyl ether,  $CH_3OCH_3$ , is \_\_\_\_\_.
- Which of the following choices represent(s) a pair of resonance forms?



# 20 4 Practice Problems Chemistry Answers

**Laurie Sorge**



## 20 4 Practice Problems Chemistry Answers:

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### Table of Contents **20 4 Practice Problems Chemistry Answers**

1. Understanding the eBook **20 4 Practice Problems Chemistry Answers**
  - The Rise of Digital Reading **20 4 Practice Problems Chemistry Answers**
  - Advantages of eBooks Over Traditional Books
2. Identifying **20 4 Practice Problems Chemistry Answers**
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an **20 4 Practice Problems Chemistry Answers**
  - User-Friendly Interface
4. Exploring eBook Recommendations from **20 4 Practice Problems Chemistry Answers**
  - Personalized Recommendations
  - **20 4 Practice Problems Chemistry Answers** User Reviews and Ratings
  - **20 4 Practice Problems Chemistry Answers** and Bestseller Lists

5. Accessing 20 4 Practice Problems Chemistry Answers Free and Paid eBooks
  - 20 4 Practice Problems Chemistry Answers Public Domain eBooks
  - 20 4 Practice Problems Chemistry Answers eBook Subscription Services
  - 20 4 Practice Problems Chemistry Answers Budget-Friendly Options
6. Navigating 20 4 Practice Problems Chemistry Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - 20 4 Practice Problems Chemistry Answers Compatibility with Devices
  - 20 4 Practice Problems Chemistry Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 20 4 Practice Problems Chemistry Answers
  - Highlighting and Note-Taking 20 4 Practice Problems Chemistry Answers
  - Interactive Elements 20 4 Practice Problems Chemistry Answers
8. Staying Engaged with 20 4 Practice Problems Chemistry Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 20 4 Practice Problems Chemistry Answers
9. Balancing eBooks and Physical Books 20 4 Practice Problems Chemistry Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 20 4 Practice Problems Chemistry Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 20 4 Practice Problems Chemistry Answers
  - Setting Reading Goals 20 4 Practice Problems Chemistry Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 20 4 Practice Problems Chemistry Answers
  - Fact-Checking eBook Content of 20 4 Practice Problems Chemistry Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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