

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your notebook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

atmosphere	integrity gap	ozone	ozone hole
stratosphere	stratosphere	ultraviolet radiation	

Earth's (90) \_\_\_\_\_ is made up of several layers. The air we breathe makes up the lowest layer. This layer is called the (80) \_\_\_\_\_. The next layer up is called the (82) \_\_\_\_\_. This layer contains a protective (81) \_\_\_\_\_ layer.

Closest to us is the (83) \_\_\_\_\_, is made by ultraviolet radiation in the upper part of the stratosphere. The ozone hole is a gap around Earth, with a (84) \_\_\_\_\_ without ozone, you are more likely to get a sunburn or possibly skin cancer. The opening of the ozone layer called the (85) \_\_\_\_\_. As mentioned, however, without ozone all organisms on Earth are called in danger from too much radiation.

In your notebook, read about chlorofluorocarbons.

Use each statement below, write in or follow.

- \_\_\_\_\_ (8) CFC is another name for a chlorofluorocarbon.
- \_\_\_\_\_ (9) CFCs are made up of carbon, fluorine, and chlorine.
- \_\_\_\_\_ (10) All CFCs are synthetic chemicals.
- \_\_\_\_\_ (11) CFCs usually react easily with other chemicals.
- \_\_\_\_\_ (12) CFCs were designed as replacements for toxic refrigerants.

# 1stoichiometry Study Guide For Mastery Content Answers

**Chao Zhang**



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