

Describing Chemical Reactions

Introduction and Purpose

In this experiment you will observe examples of the five basic types of chemical reactions. You will learn to write balanced equations, including the role of energy, to effectively communicate the chemistry of the reactions.

Procedure

Carry out the reactions using the approximate quantities of reagents suggested. Unless otherwise stated, use test tubes. When heating reagents in test tubes, slant the test tube so that the opening is pointed *away* from people. Heat the test tube at the surface of the material and work down towards the bottom of the tube. Discard solutions down the drain, wash and rinse your glassware. Discard solid waste in the waste cans on the lab tables.

In the data section you will balance the equation, write the word equation and record your observations.

A. Combination reactions:

1. Grasp a strip of magnesium ribbon in crucible tongs and ignite it in the burner flame. Hold it over a watch glass. Do not look directly at the flame! Add a few drops of distilled H_2O to the ash. Stir with a stirring rod and place a drop of the solution on red litmus paper. Red litmus turning blue is evidence for the presence of a base.
2. Heat a piece of copper metal strongly in the Bunsen burner flame for about 30 s. Remove the copper from the flame and note the change in appearance. Discard the product in the solid waste can.

B. Decomposition reactions:

1. Place about 1 scoopful of solid calcium carbonate CaCO_3 into a dry test tube and heat strongly for 2 minutes. Toward the end of the heating, light a wood splint and insert the flaming splint into the mouth of the test tube. Note what happens to the splint.

C. Single replacement reactions:

1. Place a strip of copper in a test tube with enough 0.1M AgNO_3 to cover it. Set this test tube aside, then observe the surface of the metal after 5-10 minutes.
2. Place a couple of pieces of mossy zinc metal in a test tube approximately 1/4 full of 3M HCl . Place a stopper loosely in the tube. After a few minutes, light a wood splint and insert the flaming splint into the mouth of the test tube. Hold the test tube in your hand to feel if the temperature has changed.

1section Review Describing Chemical Reactions Answer Key

Robert Spilman Turner



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Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

Patai's 1992 Guide to the Chemistry of Functional Groups Saul Patai, 1992-04-08 Patai's 1992 Guide to the Chemistry of Functional Groups Saul Patai The Hebrew University of Jerusalem Israel Ever since the publication of the first volume of The Chemistry of Functional Groups in 1964 the Patai series has acted as an essential reference source to many researchers By the end of 1991 the series consisted of 50 titles bound in 73 volumes containing nearly 900 chapters written by over 1250 authors The aim of this Guide as was that of the previous edition is to present sufficient material on each of the published chapters to allow the researcher to decide whether these chapters are relevant and useful for his or her purpose and thus worth pursuing in full For those who are familiar with only selected volumes from the series the cross referencing between complementary and related chapters from different volumes will be invaluable The Guide is fully indexed by both subject and author thus making it an essential reference tool for all organic chemists

[Ion-Selective Electrode Reviews](#) J. D. R. Thomas, 2017-05-04 Ion Selective Electrode Reviews Volume 6 covers new fields of application for coated wire ion selective electrodes ISEs The book discusses the practical suggestions for testing automatic equipment based on ISE the amperometric and potentiometric determinations with immobilized enzymes and microorganisms and the drug type substances analysis with membrane electrodes The text also describes coated wire ISEs and the in vivo application of ion sensitive field effect transistors including a summary of laboratory research and probable future clinical detectors

Paint, Oil and Chemical Review ..., 1912

Holt Chemistry Ralph Thomas Myers, 2004

Si Silicon, 2013-11-11 This volume concludes the coverage of silicon carbide SiC begun in Silicon Supplement Volume B 2 1984 subtitled Silicon Carbide Part I Part I described the physical properties of SiC SiC diodes molecular species in the SiC C gas phase and amorphous silicon carbon alloys The current Part II Silicon Supplement Volume B 3 1986 covers in its initial chapter the Si C phase diagram and in the final chapters the higher order systems of Si and C with additional elements through boron arranged according to the Gmelin system In between some 95% of the volume focusses on SiC beginning with its natural occurrence preparation and formation and purification continuing with its chemical analysis manufacture of specialized forms electrochemistry and chemical reactions and concluding with descriptions of its myriad applications The final applications section covering electronic devices also describes similar applications of the amorphous Si C alloys The successive chapters in this volume are often closely interrelated since it is often necessary to synthesize SiC directly in a form in which it will be applied SiC cannot be melted and cast nor rolled nor drawn nor is it easily electroplated or sintered

or purified Silicon carbide first became known to man when E G Acheson in 1891 used an electric current to heat a mixture of clay and carbon to extremely high temperatures *Watershed and Lake Processes Affecting Surface Water Acid-base Chemistry* Robert Spilman Turner,1990 **Glencoe Science** McGraw-Hill Staff,2001-06 Harcourt Science: Physical science, [grade] 3, units E and F, teacher's ed ,2000 First Australasian Conference on Heat and Mass Transfer, Monash University, May 23, 24, 25, 1973 ,1973 **The Electrical Review** ,1918 Monthly Technical Review ,1957

Encyclopedia of Glass Science, Technology, History, and Culture Pascal Richet,2021-02-05 A comprehensive and up to date encyclopedia to the fabrication nature properties uses and history of glass The Encyclopedia of Glass Science Technology History and Culture has been designed to satisfy the needs and curiosity of a broad audience interested in the most varied aspects of material that is as old as the universe As described in over 100 chapters and illustrated with 1100 figures the practical importance of glass has increased over the ages since it was first man made four millennia ago The old age glass vessels and window and stained glass now coexist with new high tech products that include for example optical fibers thin films metallic bioactive and hybrid organic inorganic glasses amorphous ices or all solid state batteries In the form of scholarly introductions the Encyclopedia chapters have been written by 151 noted experts working in 23 countries They present at a consistent level and in a self consistent manner these industrial technological scientific historical and cultural aspects Addressing the most recent fundamental advances in glass science and technology as well as rapidly developing topics such as extra terrestrial or biogenic glasses this important guide Begins with industrial glassmaking Turns to glass structure and to physical transport and chemical properties Deals with interactions with light inorganic glass families and organically related glasses Considers a variety of environmental and energy issues And concludes with a long section on the history of glass as a material from Prehistory to modern glass science The Encyclopedia of Glass Science Technology History and Culture has been written not only for glass scientists and engineers in academia and industry but also for material scientists as well as for art and industry historians It represents a must have comprehensive guide to the myriad aspects this truly outstanding state of matter *Applied Mechanics Reviews* ,1991 **Pharmaceutical Record and Weekly Market Review** P. W. Bedford,1886 *Chemical Thermodynamics of Compounds and Complexes of U, Np, Pu, Am, Tc, Se, Ni and Zr With Selected Organic Ligands* ,2005-11-04 This volume is part of the series on Chemical Thermodynamics published under the aegis of the OECD Nuclear Energy Agency It contains a critical review of the literature on thermodynamic data for compounds of complexes of oxalate citrate EDTA and iso saccharinate with uranium neptunium plutonium americium technetium selenium nickel and zirconium A review team composed of five internationally recognized experts has critically reviewed all the scientific literature containing chemical thermodynamic information for the above mentioned systems The results of this critical review carried out following the Guidelines of the OECD NEA Thermochemical Database Project have been documented in the present volume which contains tables of selected values for formation and reaction thermodynamical

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