

2.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions taking place under given conditions.
3. Apply the law of conservation of energy to chemical reactions.

Vocabulary

- | | | |
|-----------------------|---------------|----------------------------|
| 1. endothermic | 2. exothermic | 3. enthalpy |
| 4. formation reaction | 5. product | 6. thermochemical equation |

Part A Completion

Write the coefficient needed to balance each chemical equation, and write the correct physical state for each substance. Assume water is produced either as a liquid or as a gas.

- | | |
|---|-----------|
| 1. Aluminum reacts with oxygen to form aluminum oxide. _____ | 6. _____ |
| 2. Carbon disulfide reacts with oxygen to form carbon dioxide and sulfur dioxide. _____ | 7. _____ |
| 3. Hydrogen gas reacts with oxygen gas to form water. _____ | 8. _____ |
| 4. Nitrogen gas reacts with hydrogen gas to form ammonia. _____ | 9. _____ |
| 5. Ethanol reacts with oxygen to form carbon dioxide and water. _____ | 10. _____ |
| 6. Methane reacts with oxygen to form carbon dioxide and water. _____ | 11. _____ |
| 7. Propane reacts with oxygen to form carbon dioxide and water. _____ | 12. _____ |
| 8. Butane reacts with oxygen to form carbon dioxide and water. _____ | 13. _____ |
| 9. Pentane reacts with oxygen to form carbon dioxide and water. _____ | 14. _____ |
| 10. Hexane reacts with oxygen to form carbon dioxide and water. _____ | 15. _____ |
| 11. Heptane reacts with oxygen to form carbon dioxide and water. _____ | 16. _____ |
| 12. Octane reacts with oxygen to form carbon dioxide and water. _____ | 17. _____ |
| 13. Nonane reacts with oxygen to form carbon dioxide and water. _____ | 18. _____ |
| 14. Decane reacts with oxygen to form carbon dioxide and water. _____ | 19. _____ |
| 15. Undecane reacts with oxygen to form carbon dioxide and water. _____ | 20. _____ |
| 16. Dodecane reacts with oxygen to form carbon dioxide and water. _____ | 21. _____ |
| 17. Tridecane reacts with oxygen to form carbon dioxide and water. _____ | 22. _____ |
| 18. Tetradecane reacts with oxygen to form carbon dioxide and water. _____ | 23. _____ |
| 19. Pentadecane reacts with oxygen to form carbon dioxide and water. _____ | 24. _____ |
| 20. Hexadecane reacts with oxygen to form carbon dioxide and water. _____ | 25. _____ |
| 21. Heptadecane reacts with oxygen to form carbon dioxide and water. _____ | 26. _____ |
| 22. Octadecane reacts with oxygen to form carbon dioxide and water. _____ | 27. _____ |
| 23. Nonadecane reacts with oxygen to form carbon dioxide and water. _____ | 28. _____ |
| 24. Eicosane reacts with oxygen to form carbon dioxide and water. _____ | 29. _____ |
| 25. Heneicosane reacts with oxygen to form carbon dioxide and water. _____ | 30. _____ |
| 26. Docosane reacts with oxygen to form carbon dioxide and water. _____ | 31. _____ |
| 27.tricosane reacts with oxygen to form carbon dioxide and water. _____ | 32. _____ |
| 28.tricosane reacts with oxygen to form carbon dioxide and water. _____ | 33. _____ |
| 29.tricosane reacts with oxygen to form carbon dioxide and water. _____ | 34. _____ |
| 30.tricosane reacts with oxygen to form carbon dioxide and water. _____ | 35. _____ |

Part B True-False

Indicate whether each statement is true or false. If false, explain why.

1. The enthalpy of formation of a compound is the same as the enthalpy of combustion of the compound.
2. The enthalpy of formation of a compound is the same as the enthalpy of decomposition of the compound.
3. The enthalpy of formation of a compound is the same as the enthalpy of synthesis of the compound.

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S Ben Porath

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