Balancing Equations Worksheet II

Count the number of atoms, then use coefficients to balance each equation.

AlBr₃ + ___ K → ___ KBr + ___ Al

2) ____ FeO + ___ PdF₂ → ___ FeF₂ + ___ PdO

3) ____ P₄ + ____ Br₂ → ____ PBr₃

4) ____ LiCl + ___ Br₂ → ___ LiBr + ___ Cl₂

5) ____ PbBr₂ + ___ HCl → ___ HBr + ___ PbCl₂

6) ____ CoBr₃ + ____ CaSO₄ → ____ CaBr₂ + ____ Co₂(SO₄)₃

7) ____ Na₃P + ___ CaF₂ → ___ NaF + ___ Ca₃P₂

8) ____ Mn + ___ HI → ___ H₂ + ___ MnI₃

9) ____ Li₃PO₄ + ____ NaBr → ____ Na₃PO₄ + ____ LiBr

10) ____ CaF₂ + ____ Li₂SO₄ → ____ CaSO₄ + ____ LiF

11) ____ HBr + ___ Mg(OH)₂ → ___ MgBr₂ + ___ H₂O

12) ____ LiNO₃ + ____ CaBr₂ → ____ Ca(NO₃)₂ + ____ LiBr

13) ____ AgNO₃ + ___ Li → ___ LiNO₃ + ___ Ag

14) ____ Si(OH)₄ + ___ NaBr → ___ SiBr₄ + ___ NaOH

15) ____ NaCN + ___ CuCO₃ → ___ Na₂CO₃ + ___ Cu(CN)₂

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Arshad Iqbal

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Table of Contents 1chemical Reactions Practice Problems Answers 11

- 1. Understanding the eBook 1chemical Reactions Practice Problems Answers 11
 - The Rise of Digital Reading 1chemical Reactions Practice Problems Answers 11
 - Advantages of eBooks Over Traditional Books
- 2. Identifying 1chemical Reactions Practice Problems Answers 11
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1chemical Reactions Practice Problems Answers 11
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from 1chemical Reactions Practice Problems Answers 11
 - Personalized Recommendations
 - 1chemical Reactions Practice Problems Answers 11 User Reviews and Ratings
 - 1chemical Reactions Practice Problems Answers 11 and Bestseller Lists

- 5. Accessing 1chemical Reactions Practice Problems Answers 11 Free and Paid eBooks
 - 1chemical Reactions Practice Problems Answers 11 Public Domain eBooks
 - 1chemical Reactions Practice Problems Answers 11 eBook Subscription Services
 - 1chemical Reactions Practice Problems Answers 11 Budget-Friendly Options
- 6. Navigating 1chemical Reactions Practice Problems Answers 11 eBook Formats
 - o ePub, PDF, MOBI, and More
 - 1chemical Reactions Practice Problems Answers 11 Compatibility with Devices
 - 1chemical Reactions Practice Problems Answers 11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 1chemical Reactions Practice Problems Answers 11
 - Highlighting and Note-Taking 1chemical Reactions Practice Problems Answers 11
 - Interactive Elements 1chemical Reactions Practice Problems Answers 11
- 8. Staying Engaged with 1chemical Reactions Practice Problems Answers 11
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1chemical Reactions Practice Problems Answers 11
- 9. Balancing eBooks and Physical Books 1chemical Reactions Practice Problems Answers 11
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection 1chemical Reactions Practice Problems Answers 11
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 1chemical Reactions Practice Problems Answers 11
 - Setting Reading Goals 1chemical Reactions Practice Problems Answers 11
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 1chemical Reactions Practice Problems Answers 11
 - Fact-Checking eBook Content of 1chemical Reactions Practice Problems Answers 11
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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