## Chemical Reaction:

$$2N\alpha_2IICO_3 + II_2SO_4 = N\alpha_2SO_4 + 2II_2O + 2CO_3$$

#### Procedure:

240

## For Carbonates (CO3):

- Take 50ml of water sample in conical flask.
- Add 2-3 drops of phenolphthalein.
- If pink colour appears, carbonates are present.
- Titrate it against H<sub>2</sub>SO<sub>4</sub> to a colourless end point.
- Take readings for volume of H<sub>2</sub>SO<sub>4</sub>.
- Calculate the carbonates with the given formula/

#### For Bicarbonates (HCO3):

- Take 50ml of water sample in conical flask.
- Add 2-3 drops of methyl orange.
- If orange colour appears, bicarbonates are present.
- Titrate it against H<sub>2</sub>SO<sub>4</sub> to a pinkish yellow end point.
- Calculate the carbonates with the given formula/
- Take 3 readings for volume of H<sub>2</sub>SO<sub>4</sub>.

# **Amount Of Carbonate In An Ore Sample**

**Sebastian Brünink** 

#### **Amount Of Carbonate In An Ore Sample:**

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