

CHEM 111 Name: \_\_\_\_\_ Section: \_\_\_\_\_

Experiment #15: Activity Series  
Pre-Lab Questions

1. What cation will be in solution when each of the following compounds are dissolved in water?

- a)  $\text{Ca}(\text{NO}_3)_2$       e)  $\text{Mg}(\text{NO}_3)_2$   
b)  $\text{CuSO}_4$       f)  $\text{SrCl}_2$   
c)  $\text{FeSO}_4$       g)  $\text{Zn}(\text{NO}_3)_2$   
d)  $\text{Fe}(\text{NO}_3)_3$       h)  $\text{HCl}$

2. First, predict which of the following reactions will take place. For the reactions that will take place, complete and balance the reaction equations. Then indicate which molecule was oxidized and which was reduced.



3. Use the following information to arrange the metals A, B, and C in order of decreasing reactivity starting with the most reactive (1) and ending with the least reactive (3).



(1) \_\_\_\_\_

(2) \_\_\_\_\_

(3) \_\_\_\_\_

# Activity Series Pre Lab Answers 15

**Seung-Beom Hong,M. Bazlur**

**Rashid,Lory Z. Santiago-Vázquez**

## Activity Series Pre Lab Answers 15:

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