# 11-1 Practice Problems

 Lead will react with Hydrochloric acid to produce Lead (II) Chloride and Hydrogen gas. How many moles of Hydrochloric acid are needed to completely react with 0.36-mol of lead? 6. Carbon will react with Zinc Oxide to produce Zinc and Carbon Dioxide. How many moles of carbon dioxide will be produced if 0.38-mol of ZnO is completely reacted?

 How many moles of HNO<sub>3</sub> will be produced when 0.51-mol of N<sub>2</sub>O<sub>5</sub> reacts according to the following equation? N<sub>2</sub>O<sub>5</sub> + H<sub>2</sub>O → 2HNO<sub>3</sub>  How many moles of oxygen will be needed to react with 0.38-mol of C<sub>3</sub>H<sub>8</sub> according to the following equation?
 C<sub>3</sub>H<sub>8</sub> + 5O<sub>2</sub> → 3CO<sub>2</sub> + 4H<sub>2</sub>O

3. Iron will react with Oxygen Gas to produce Fe<sub>2</sub>O<sub>3</sub>. How many moles of Fe<sub>2</sub>O<sub>3</sub> will be produced if 0.18-mol of Fe reacts?  Nitrogen gas can react with Hydrogen gas to produce ammonia. How many moles of Nitrogen will be needed to produce 0.48-mol of NH<sub>3</sub>?

 How many moles of NaBr will be produced when 0.69-mol of Bromine reacts according to the following equation?
 Br<sub>2</sub> + 2NaI → 2NaBr + I<sub>2</sub>  How many moles of hydrogen will be produced if 0.44-mol of CaH<sub>2</sub> reacts according to the following equation? CaH<sub>2</sub> + 2H<sub>2</sub>O → Ca(OH)<sub>2</sub> + 2H<sub>2</sub>

5. Phosphorus will react with Bromine to produce Phosphorus Tribromide. How many moles of Phosphorus Tribromide will be produced if 0.78-mol of Bromine is reacted?  How many moles of water will be produced if 2.35-mol of oxygen reacts according to the following equation?
 2C<sub>6</sub>H<sub>6</sub> + 15O<sub>2</sub> → 12CO<sub>2</sub> + 6H<sub>2</sub>O

# **Adapted From Prentice Hall 11 1 Practice Problems Answers**

**Daniel F McAuley** 

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