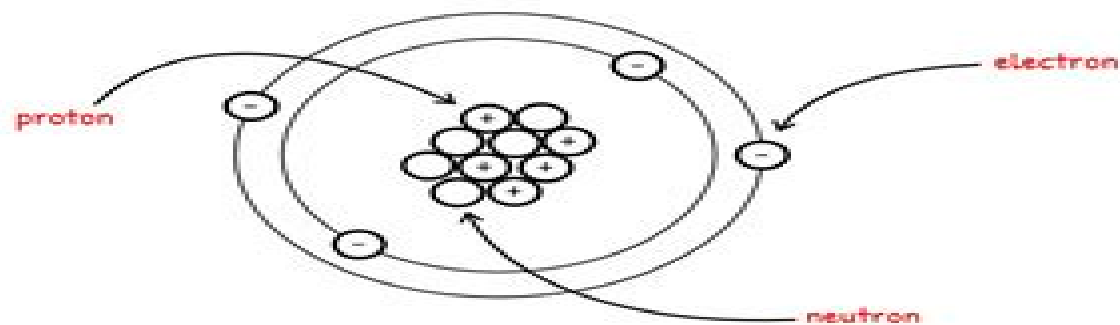


Name answer key

Date _____

Atomic Structure

Label the parts of an atom on the diagram below, and answer the questions!



1. What type of charge does a proton have?
Positive (+1)
2. What type of charge does a neutron have?
Neutral (0)
3. What type of charge does an electron have?
Negative (-1)
4. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons
5. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35
6. What is the atomic number of the atom in the diagram above?
5
7. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
8. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number
9. How many electrons are in the nucleus of an atom with an atomic number of 20?
20
10. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass) 30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30
11. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number = 6.9% (Li-Lithium has an atomic number of 3 which means it has 3 protons.
12. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons = 11 (atomic mass: 36 - 25:atomic number)

4 Atomic Structure Practice Problems Answers

Arshad Iqbal



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MCQ Chapter 20 Nitrogen and Sulfur MCQ Chapter 21 Organic and Nitrogen Compounds MCQ Chapter 22 Periodicity MCQ Chapter 23 Polymerization MCQ Chapter 24 Rates of Reaction MCQ Chapter 25 Reaction Kinetics MCQ Chapter 26 Redox Reactions and Electrolysis MCQ Chapter 27 States of Matter MCQ Chapter 28 Transition Elements MCQ The Alcohols and Esters MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Introduction to alcohols and alcohols reactions The Atomic Structure and Theory MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Atom facts elements and atoms number of nucleons protons electrons and neutrons The Benzene Chemical Compound MCQ PDF e Book Chapter 3 practice test to solve MCQ questions on Introduction to benzene arenes reaction phenol and properties and reactions of phenol The Carbonyl Compounds MCQ PDF e Book Chapter 4 practice test to solve MCQ questions on Introduction to carbonyl compounds aldehydes and ketone testing nucleophilic addition with HCN preparation of aldehydes and ketone reduction of aldehydes and ketone The Carboxylic Acids and Acyl Compounds MCQ PDF e Book Chapter 5 practice test to solve MCQ questions on Acidity of carboxylic acids acyl chlorides ethanoic acid and reactions to form tri iodomethane The Chemical Bonding MCQ PDF e Book Chapter 6 practice test to solve MCQ questions on Chemical bonding types chemical bonding electron pair bond angle bond energy bond length bonding and physical properties bonding energy repulsion theory covalent bonding covalent bonds double covalent bonds triple covalent bonds electron pair repulsion and bond angles electron pair repulsion theory enthalpy change of vaporization intermolecular forces ionic bonding ionic bonds and covalent bonds ionic bonds metallic bonding metallic bonding and delocalized electrons number of electrons sigma bonds and pi bonds sigma bonds pi bonds s orbital and p orbital Van der Waals forces and contact points The Chemistry of Life MCQ PDF e Book Chapter 7 practice test to solve MCQ questions on Introduction to chemistry enzyme specificity enzymes reintroducing amino acids and proteins The Electrode Potential MCQ PDF e Book Chapter 8 practice test to solve MCQ questions on Electrode potential cells and batteries E Plimsoll values electrolysis process measuring standard electrode potential quantitative electrolysis redox and oxidation The Electrons in Atoms MCQ PDF e Book Chapter 9 practice test to solve MCQ questions on Electronic configurations electronic structure evidence ionization energy periodic table simple electronic structure sub shells and atomic orbitals The Enthalpy Change MCQ PDF e Book Chapter 10 practice test to solve MCQ questions on Standard enthalpy changes bond energies enthalpies Hess law introduction to energy changes measuring enthalpy changes The Equilibrium MCQ PDF e Book Chapter 11 practice test to solve MCQ questions on Equilibrium constant expression equilibrium position acid base equilibria chemical industry equilibria ethanoic acid gas reactions equilibria and reversible reactions The Group IV MCQ PDF e Book Chapter 12 practice test to solve MCQ questions on Introduction to group IV metallic character of group IV elements ceramic silicon oxide covalent bonds properties variation in group IV relative stability of oxidation states and tetra chlorides The Groups II and VII MCQ PDF e Book Chapter 13 practice test to solve MCQ questions on Atomic number of group II metals covalent bonds density of group II elements

disproportionation fluorine group II elements and reactions group VII elements and reactions halogens and compounds ionic bonds melting points of group II elements metallic radii of group II elements periodic table elements physical properties of group II elements physical properties of group VII elements reaction of group II elements with oxygen reactions of group II elements reactions of group VII elements thermal decomposition of carbonates and nitrates thermal decomposition of group II carbonates thermal decomposition of group II nitrates uses of group II elements uses of group II metals uses of halogens and their compounds The Halogenoalkanes MCQ PDF e Book Chapter 14 practice test to solve MCQ questions on Halogenoalkanes uses of halogenoalkanes elimination reactions nucleophilic substitution in halogenoalkanes and nucleophilic substitution reactions The Hydrocarbons MCQ PDF e Book Chapter 15 practice test to solve MCQ questions on Introduction to alkanes sources of alkanes addition reactions of alkenes alkane reaction alkenes and formulas The Introduction to Organic Chemistry MCQ PDF e Book Chapter 16 practice test to solve MCQ questions on Organic chemistry functional groups organic reactions naming organic compounds stereoisomerism structural isomerism and types of organic reactions The Ionic Equilibria MCQ PDF e Book Chapter 17 practice test to solve MCQ questions on Introduction to ionic equilibria buffer solutions equilibrium and solubility indicators and acid base titrations pH calculations and weak acids The Lattice Energy MCQ PDF e Book Chapter 18 practice test to solve MCQ questions on Introduction to lattice energy ion polarization lattice energy value atomization and electron affinity Born Haber cycle and enthalpy changes in solution The Moles and Equations MCQ PDF e Book Chapter 19 practice test to solve MCQ questions on Amount of substance atoms molecules mass chemical formula and equations gas volumes mole calculations relative atomic mass solutions and concentrations The Nitrogen and Sulfur MCQ PDF e Book Chapter 20 practice test to solve MCQ questions on Nitrogen gas nitrogen and its compounds nitrogen and gas properties ammonia ammonium compounds environmental problems caused by nitrogen compounds and nitrate fertilizers sulfur and oxides sulfuric acid and properties and uses of sulfuric acid The Organic and Nitrogen Compounds MCQ PDF e Book Chapter 21 practice test to solve MCQ questions on Amides in chemistry amines amino acids peptides and proteins The Periodicity MCQ PDF e Book Chapter 22 practice test to solve MCQ questions on Acidic oxides basic oxides aluminum oxide balancing equation period 3 chlorides balancing equations reactions with chlorine balancing equations reactions with oxygen bonding nature of period 3 oxides chemical properties of chlorine chemical properties of oxygen chemical properties periodicity chemistry periodic table chemistry oxides chlorides of period 3 elements electrical conductivity in period 3 oxides electronegativity of period 3 oxides ionic bonds molecular structures of period 3 oxides oxidation number of oxides oxidation numbers oxides and hydroxides of period 3 elements oxides of period 3 elements period III chlorides periodic table electronegativity physical properties periodicity reaction of sodium and magnesium with water and relative melting point of period 3 oxides The Polymerization MCQ PDF e Book Chapter 23 practice test to solve MCQ questions on Types of polymerization polyamides polyesters and polymer deductions The Rates of Reaction MCQ PDF e Book

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Table of Contents 4 Atomic Structure Practice Problems Answers

1. Understanding the eBook 4 Atomic Structure Practice Problems Answers
 - The Rise of Digital Reading 4 Atomic Structure Practice Problems Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 4 Atomic Structure Practice Problems Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 4 Atomic Structure Practice Problems Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 4 Atomic Structure Practice Problems Answers
 - Personalized Recommendations
 - 4 Atomic Structure Practice Problems Answers User Reviews and Ratings
 - 4 Atomic Structure Practice Problems Answers and Bestseller Lists
5. Accessing 4 Atomic Structure Practice Problems Answers Free and Paid eBooks

- 4 Atomic Structure Practice Problems Answers Public Domain eBooks
 - 4 Atomic Structure Practice Problems Answers eBook Subscription Services
 - 4 Atomic Structure Practice Problems Answers Budget-Friendly Options
6. Navigating 4 Atomic Structure Practice Problems Answers eBook Formats
- ePub, PDF, MOBI, and More
 - 4 Atomic Structure Practice Problems Answers Compatibility with Devices
 - 4 Atomic Structure Practice Problems Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 4 Atomic Structure Practice Problems Answers
 - Highlighting and Note-Taking 4 Atomic Structure Practice Problems Answers
 - Interactive Elements 4 Atomic Structure Practice Problems Answers
8. Staying Engaged with 4 Atomic Structure Practice Problems Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 4 Atomic Structure Practice Problems Answers
9. Balancing eBooks and Physical Books 4 Atomic Structure Practice Problems Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 4 Atomic Structure Practice Problems Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 4 Atomic Structure Practice Problems Answers
- Setting Reading Goals 4 Atomic Structure Practice Problems Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 4 Atomic Structure Practice Problems Answers
- Fact-Checking eBook Content of 4 Atomic Structure Practice Problems Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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