



Name: _____
 Period: _____

KEY

Nuclear Radiation Worksheet

- What determines whether an atom is stable or unstable?
Its overall size and a difference between proton # and neutron #
 - How does an unstable atom gain stability?
It undergoes radioactive decay
 - What part of an atom undergoes change during radioactive decay?
The nucleus!

- Which of the human senses can detect radiation?
Our eyes can detect colors, but we have no sense to detect alpha, beta, or gamma radiation

- Explain the difference between an isotope and a radioactive isotope.

An isotope has a different # of protons / neutrons. Radioactive isotope is larger, unstable, and has a different # of p

- Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	<i>4 protons 4 neutrons</i>	Positive (+)	<i>Stopped by paper or skin</i>
Beta	<i>0 No Mass (electron)</i>	Negative (-)	<i>Stopped by sheets of tin foil or thick paper</i>
Gamma	<i>0 No mass</i>	Neutral	<i>Stopped by several inches of lead or feet of concrete</i>

- How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	<i>-4</i>	<i>-2</i>
Beta	<i>No change</i>	<i>+1</i>
Gamma	<i>No change</i>	<i>No change</i>

- Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



- Complete the following nuclear equations:



Answer Key To Nuclear Radiation

Xiang Xie



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