Ionic and Covalent Naming Practice Answers

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

1)	Na ₂ CO ₃	sodium carbonate	
2)	P4O10	tetraphosphorus decaoxide	
3)	N ₂ O ₅	dinitrogen pentoxide	
4)	FeSO ₄	iron (II) sulfate	
5)	SiO2	silicon dioxide	
6)	FeCl _a	iron (III) chloride	
7)	CoBr ₂	cobalt (II) bromide	
8)	B ₂ H ₄	diboron tetrahydride	
9)	CO	carbon monoxide	
10)	PCI ₃	phosphorus trichloride	

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

11)	dinitrogen trioxide	N ₂ O ₃
12)	nitrogen dioxide	NO ₂
13)	sulfur pentoxide	SOs
14)	lithium acetate	LIC2H3O2
15)	phosphorus trifluoride	PF ₃
16)	vanadium (V) oxide	V ₂ O ₅
17)	aluminum hydroxide	Al(OH)3
18)	zinc sulfide	ZnS
19)	silicon tetrafluoride	SiF.
20)	silver phosphate	Ag ₃ PO ₄

Answer Key For Chapter Assessment Ionic Compounds

SJ Ball

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