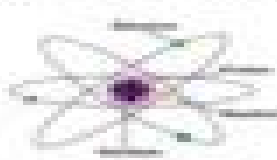


Chemical Bonding WebQuest

Directions: Use the link to complete the WebQuest!



<https://www.youtube.com/watch?v=5m10w0w3p0g> Baseline video

The _____ around us is made up of tiny units of matter called _____. How these _____ stick together to form _____ is called chemical bonding.

About Atoms

Each _____ has its own unique _____ made up of a specific number of protons in its nucleus called the atomic _____. Each atom also has the same number of _____ as it has _____.

Electron Shells

The electrons _____ around the _____ of the atom. They stay in layers called shells. Each _____ can only contain a _____ number of electrons: the first layer can hold _____ electrons, the second layer _____ electrons, the _____ layer _____ electrons, etc.

The Outer Shell

All _____ would like to have a full _____ shell, but the only elements to naturally have a full outer shell are the noble gases to the right of the _____ table. As a result, when atoms without full _____ shells come into contact with other atoms, they _____ to want to give up or gain _____.

Valence Electrons

The _____ electrons are the number of electrons in an outer shell of an atom that can participate in forming chemical bonds with other atoms.

Atoms with a _____ empty outer shell will want to give up _____. For example, if an atom has 1 electron out of a possible _____ in its outer shell, it will want to give up that electron so its outer _____ is now full.

Answer Key To Chemical Bonding Webquest

Rachel Turoscy



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