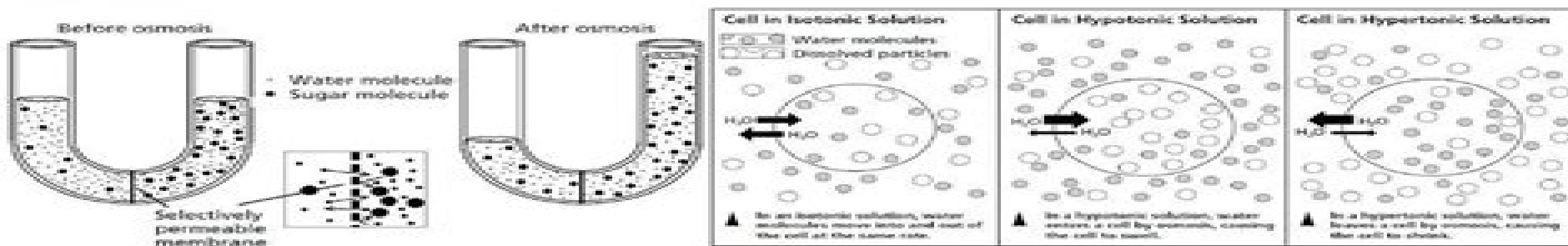


1. As a relative concentration of solutes in two solutions increases, of necessity the concentration of water in the same two solutions **INCREASES**. Solution A has a lower concentration of **SOLUTE** than does Solution C; Solution A is also hypertonic to Solution C.

3. If you wanted to make Solution A isotonic to Solution B, you could add water to Solution B or you could add solute to Solution A. If you took all three solutions, put them into a large container and mixed them thoroughly, then redistributed the solution among three containers, Solution A would be **ISOTONIC** to Solution B. Solution A would also be **ISOTONIC** to Solution C, and Solution C would be **ISOTONIC** to Solution B.

**Part III – Look at the solutions and cells illustrated above and fill in the blanks.**

1. Because the cytoplasm of the plant and the animal cell have equal concentrations of solutes, we can say their cytoplasm is **ISOTONIC** to each other. If we put both the plant and the animal cells into Solution A, we would expect **no change** in the cells, because Solution A is **ISOTONIC** to the cytoplasm of each cell.
2. Let's put both cells into Solution B. Because Solution B is hypertonic to the cytoplasm of the cells, we would expect water to **LEAVE** the cells through the process of **OSMOSIS**. This would result in the cytoplasm of both cells shrinking.
3. Now we'll put both the plant and animal cell into Solution C, which, because it contains no solutes at all, is **HYPOTONIC** to the cytoplasm of both cells. **WATER** will enter both cells through osmosis. The animal cell is likely to **BURST**, unfortunately. The plant cell, however, is protected from this because of the presence of its **CELL WALL**.



Refer to the U-tube pictures above when answering the questions below.

1. Why did the number of water molecules on each side of the membrane change, whereas the number of sugar molecules stayed the same? **WATER MOLECULES ARE SMALL ENOUGH TO PASS THROUGH THE PORES OF THE MEMBRANE, HOWEVER, THE SUGAR MOLECULES ARE NOT.**
2. How does the plasma membrane of a cell compare with the membrane in the U-shaped tube? **THEY ARE BOTH SELECTIVELY PERMEABLE BASED ON SIZE OF PARTICLES**

# Answer Key For Osmosis And Tonicity

**Jane B. Taylor**



## **Answer Key For Osmosis And Tonicity:**

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