



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

- Chemical formulas can have three components.
 - The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
 - The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
 - The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

| 1) NaOH | |
|---------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 4) Li ₂ SO ₄ | |
|------------------------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 7) H ₂ COCH ₃ | |
|-------------------------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 10) NH ₄ C ₂ H ₃ O ₂ | |
|--|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 13) Ca ₃ (PO ₄) ₂ | |
|---|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 2) HNO ₃ | |
|---------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 5) NH ₄ Cl | |
|-----------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 8) Mg(OH) ₂ | |
|------------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 11) NaC ₂ H ₃ O ₂ | |
|--|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 14) Fe ₂ (CO ₃) ₃ | |
|---|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 3) U ₂ O | |
|---------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 6) CaClO ₄ | |
|-----------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 9) Al(OH) ₃ | |
|------------------------|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 12) (NH ₄) ₃ PO ₄ | |
|---|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

| 15) Al ₂ (SO ₄) ₃ | |
|---|------------|
| Element | # of Atoms |
| | |
| | |
| Total | |

Answer Key For Counting Atoms In Compound

RJ Alexander



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