

Practice Problems: Moles-Grains & Grains-Moles

1. How many grams of HCl are used by the reaction of 2.50 moles of magnesium?



G: 2.5 moles Mg
U: g HCl

$$\frac{2.5 \text{ moles Mg}}{1 \text{ mol Mg}} \times \frac{2 \text{ mol HCl}}{1 \text{ mol HCl}} \times \frac{36.46 \text{ g HCl}}{1 \text{ mol HCl}} = 182.3 \text{ g HCl}$$

2. How many grams of
- $\text{Ca}_3(\text{PO}_4)_2$
- would be made with 3.20 moles of
- $\text{Ca}(\text{OH})_2$
- ?



G: 3.2 mol $\text{Ca}(\text{OH})_2$
U: g $\text{Ca}_3(\text{PO}_4)_2$

$$\frac{3.2 \text{ moles Ca}(\text{OH})_2}{3 \text{ mol Ca}(\text{OH})_2} \times \frac{1 \text{ mol Ca}_3(\text{PO}_4)_2}{1 \text{ mol Ca}_3(\text{PO}_4)_2} \times \frac{310.18 \text{ g Ca}_3(\text{PO}_4)_2}{1 \text{ mol Ca}_3(\text{PO}_4)_2} = 330.8 \text{ g Ca}_3(\text{PO}_4)_2$$

3. How many moles of
- CO_2
- are produced when 25.0 grams of
- C_2H_2
- burn?



G: 25.0 g C_2H_2
U: mol CO_2

$$\frac{25.0 \text{ g C}_2\text{H}_2}{26.04 \text{ g C}_2\text{H}_2} \times \frac{1 \text{ mol C}_2\text{H}_2}{1 \text{ mol C}_2\text{H}_2} \times \frac{4 \text{ mol CO}_2}{2 \text{ mol C}_2\text{H}_2} = 1.92 \text{ mol CO}_2$$

4. How many moles of iron are produced with 25.0 grams of magnesium?



G: 25.0 g Mg
U: moles Fe

$$\frac{25.0 \text{ g Mg}}{24.30 \text{ g Mg}} \times \frac{1 \text{ mol Mg}}{1 \text{ mol Mg}} \times \frac{2 \text{ mol Fe}}{3 \text{ mol Mg}} = 0.686 \text{ mol Fe}$$

5. How many grams of SrO would be produced from 1.75 moles of
- O_2
- ?



G: 1.75 moles O_2
U: g SrO

$$\frac{1.75 \text{ mol O}_2}{1 \text{ mol O}_2} \times \frac{2 \text{ mol SrO}}{1 \text{ mol SrO}} \times \frac{103.62 \text{ g SrO}}{1 \text{ mol SrO}} = 362.7 \text{ g SrO}$$

Answer Guide For Stoichiometry Calculations

Clyde Metz



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Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

The Complete Idiot's Guide to Chemistry Ian Guch, 2003. Guch covers all the elements, the Periodic Table, ionic and covalent compounds, chemical reactions, acids and bases, and much more.

Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01. Study Guide to Accompany Calculus for the Management, Life, and Social Sciences.

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13. This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed, step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid-base reactions, which often persists through working with acid-base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14. Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd-numbered exercises.

Chemistry for Nerds Guide Book: Chemistry, Science, Nerd, Geek, Textbook, Guidebook, Study Guide, Educational, STEM, Science Gift Matt Kingsley, Calling all

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Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

The Absolute, Ultimate Guide to Lehninger Principles of Biochemistry 4e Marcy Osgood, Karen Ocorr, 2005 This undergraduate textbook describes the structure and function of the major classes of cellular constituents and explains the physical chemical and biological context in which each biomolecule reaction and pathway operates The fourth edition adds a chapter on the regulation of metabolism reflects recent advances and incorporates new experimental methodologies and an expanded and redesigned treatment of reaction mechanisms Annotation 2004 Book News Inc Portland OR booknews.com

A Visual Analogy Guide to Chemistry, 2e Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

Know Your 'O' Level Chemistry - A Study Guide, Cambridge IGCSE® Chemistry Revision Guide Roger Norris, 2015-10-08 The Cambridge IGCSE Chemistry Revision Guide supports students through their course containing specifically designed features to help students apply their knowledge as they prepare for assessment This Revision Guide offers support for students as they prepare for their Cambridge IGCSE Chemistry 0620 exams Containing up to date material that matches the syllabus for examination from 2016 and packed full of guidance such as Worked Examples Tips and Progress Check questions throughout to help students to hone their revision and exam technique and avoid common

mistakes These features have been specifically designed to help students apply their knowledge in exams Written in a clear and straightforward tone this Revision Guide is perfect for international learners *Study Guide for Introductory Chemistry*
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