

7. Alkanols can be manufactured from alkenes by the initial reaction of alkenes with
- bromine in tetrachloromethane.
  - concentrated tetraoxosulphate(VI) acid.
  - aqueous potassium tetraoxomanganate(VII).
  - sodium hydroxide solution.
8. Which of the following statements about the standard hydrogen electrode is **not** correct?
- The hydrogen gas is at a pressure of 1 atmosphere
  - A solution containing 1 mol dm<sup>-3</sup> of H<sup>+</sup><sub>(aq)</sub> ions is used
  - A platinum electrode is used
  - The temperature is kept at 20 °C
9. If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be? [O = 16.0, M = 120]
- MO
  - MO<sub>2</sub>
  - MO<sub>3</sub>
  - MO<sub>4</sub>
10. The products of the electrolysis of dilute sodium chloride using carbon electrodes are
- |    | <u>Anode</u> | <u>Cathode</u> |
|----|--------------|----------------|
| A. | chlorine     | sodium.        |
| B. | oxygen       | hydrogen.      |
| C. | chlorine     | hydrogen.      |
| D. | hydrogen     | oxygen.        |
11. Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.
- 12 C
  - 120 C
  - 360 C
  - 720 C
12. Oxochlorate(I) acid is used as a bleaching agent because it is
- a weak acid.
  - a reducing agent.
  - an oxidizing agent.
  - a strong acid.
13. The IUPAC name for CH<sub>3</sub>CHCH<sub>3</sub>CHClCH<sub>3</sub>CH<sub>2</sub>CH<sub>3</sub> is
- 2, 4 - dimethyl - 3 - chlorohexane.
  - 3, 5 - dimethyl - 4 - chlorohexane.
  - 4 - chloro - 3, 5 - dimethylhexane.
  - 3 - chloro - 2, 4 - dimethylhexane.
14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn
- acidified K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> solution colourless.
  - acidified KMnO<sub>4</sub> solution colourless.
  - Fe(NO<sub>3</sub>)<sub>2</sub> solution green.
  - Pb(NO<sub>3</sub>)<sub>2</sub> paper black.
15. If 5.0 g of marble reacts with 25.0 cm<sup>3</sup> hydrochloric acid, which of the following combinations has the fastest reaction rate?
- Marble chips and 2.0 mol dm<sup>-3</sup> HCl<sub>(aq)</sub>
  - Marble chips and 2.5 mol dm<sup>-3</sup> HCl<sub>(aq)</sub>
  - Powdered marble and 2.5 mol dm<sup>-3</sup> HCl<sub>(aq)</sub>
  - Powdered marble and 2.0 mol dm<sup>-3</sup> HCl<sub>(aq)</sub>

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