#### Six Types of Chemical Reaction Worksheet Answers

Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

-	AT THE PARTY OF TH	
as)	2 NaBr + 1 Ca(OH) <sub>2</sub> → 1 CaBr <sub>2</sub> + 2 NaOH	Type of reaction: double displacement
(D)	2 NH <sub>3</sub> + 1 H <sub>2</sub> SO <sub>4</sub> → 1 (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	Type of reaction: synthesis
c)	4 C <sub>2</sub> H <sub>9</sub> O + 27 O <sub>2</sub> → 20 CO <sub>2</sub> + 18 H <sub>2</sub> O	Type of reaction: combustion
a)	3 Pb + 2 H₂PO₁ → 3 H₂ + 1 Pb₂(PO₂)₂	Type of reaction: single displacement
(0)	1 U <sub>2</sub> N + 3 NH <sub>4</sub> NO <sub>3</sub> → 3 LINO <sub>3</sub> + 1 (NH <sub>4</sub> ) <sub>2</sub> N	Type of reaction: double displacement
7)	3 HBr + 1 Al(OH)3 -> 3 H2O + 1 AlBrs	Type of reaction: acid-base (double displacement)
-90	Na <sub>5</sub> PO <sub>4</sub> + 3 KOH → 3 NaOH + K <sub>5</sub> PO <sub>4</sub>	Reaction Typedouble displacement
h)	MgCl <sub>2</sub> + Li <sub>2</sub> CO <sub>3</sub> → MgCO <sub>3</sub> + 2 LiCl	Reaction Typedouble displacement
0	C <sub>6</sub> H <sub>12</sub> + 9 O <sub>2</sub> → 6 CO <sub>2</sub> + 6 H <sub>2</sub> O	Reaction Typecomplete combustion
10	Pb + FeSO₄ → PbSO₄ + Fe	Reaction Typesingle displacement
(k)	CaCO₂ → CaO + CO₂	Reaction Typedecomposition
-10	P4 + 3 O2 → 2 P2O3	Reaction Typesynthesis
m	) 2 RbNO <sub>5</sub> + BeF <sub>2</sub> → Be(NO <sub>5</sub> ) <sub>2</sub> + 2 RbF	Reaction Typedouble displacement
in)	2 AgNO <sub>5</sub> + Cu → Cu(NO <sub>5</sub> ) <sub>5</sub> + 2 Ag	Reaction Typesingle displacement
0)	C <sub>3</sub> H <sub>6</sub> O + 4 O <sub>2</sub> → 3 CO <sub>2</sub> + 3 H <sub>2</sub> O	Reaction Typecomplete combustion
p-)	2 C <sub>6</sub> H <sub>6</sub> + Fe → Fe(C <sub>6</sub> H <sub>6</sub> ) <sub>2</sub>	Reaction Typesynthesis
93	SeCle + O₂ → SeO₂ + 3Cl₂	Reaction Typesingle displacement
(1)	$2 \text{ Mgl}_2 + \text{Mn}(SO_3)_2 \rightarrow 2 \text{ MgSO}_3 + \text{Mnl}_4$	Reaction Typedouble displacement
5)	○3 → ○ + ○3	Reaction Typedecomposition
t)	2 NO₂ → 2 O₂ + N₂	Reaction Typedecomposition

# **Answer Classifying Chemical Reaction**

Raffaela Di Napoli

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