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AP Biology

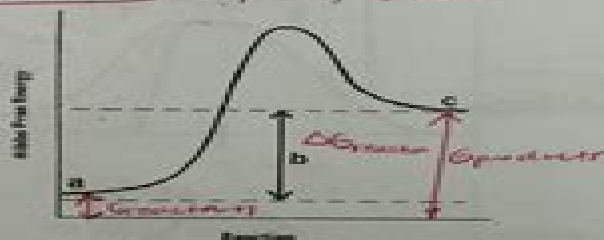
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REVIEW
Chapter 8 Metabolism/Enzymes

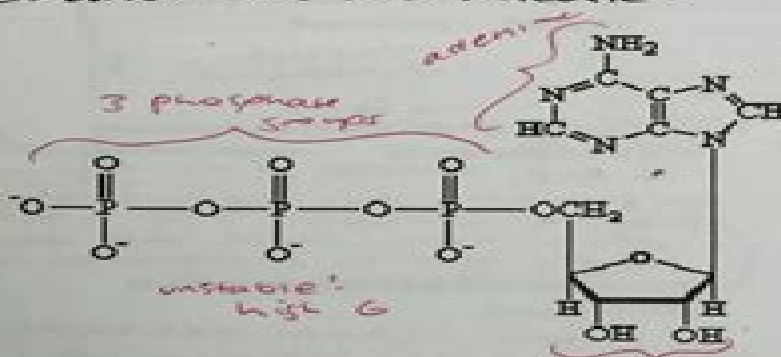
Activity 1: Thermodynamics

1. For each of the following pairs choose the scenario with lower entropy = lower S, more organized, less random
- Hydrolysis / Dehydration Synthesis
 - 3 molecules of H_2O / 6 hydrogen and 3 oxygen atoms
 - Potential energy / Heat

- (2.1) Label a-c on the graph of the reaction below.
(2.2) Is this reaction exergonic or endergonic?
(2.3) Is the reaction spontaneous or nonspontaneous? uphill, $+\Delta G$



3. Label this ATP molecule.



Activity 2: Enzymes

1. Describe the induced fit model of enzyme substrate interaction. - called by entry of the substrate the change of shape of the active site of the enzyme so that it binds most closely to the substrate
2. How do enzymes speed up the rate of a chemical reaction? lower E_a so that reaction that would have occurred anyway happens faster
3. Aspirin permanently binds to the active site of the enzyme that makes the chemical responsible for inflammation and stops it from doing its job.
- What type of inhibitor is aspirin? irreversible, competitive inhibitor
 - What type of bonds exist between aspirin and the enzyme? covalent (strong)
4. Strychnine is a poison that leads to convulsions by binding to the allosteric site of a nerve receptor protein.
- What type of inhibitor is strychnine? noncompetitive, allosteric inhibitor
 - How does it prevent the enzyme-substrate complex from forming?
binding it indirectly alters the active site conformation, preventing substrate attachment.

(1)

Answer Key To Enzyme Activity

EduCart



Answer Key To Enzyme Activity:

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CliffsStudySolver: Biology Max Rechtman, 2007-05-03

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The Pharmacy Technician, 7e Perspective Press,2020-01-15 Endorsed by the American Pharmacists Association APhA

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TEXT BOOK OF BIOCHEMISTRY Prof. (Dr.) Bhoomika, Dr. Sunita S. Pachori, Abhilasha Gupta, Kuldeep Saini, Mr. Shashank Chaurasiya, 2025-09-19

The Textbook of Biochemistry provides a comprehensive understanding of the molecular basis of life systematically covering the fundamental biomolecules and their roles in maintaining biological functions. Beginning with biomolecules, the book introduces carbohydrates, lipids, nucleic acids, amino acids, and proteins, highlighting their classification, chemical nature, and biological significance. The section on bioenergetics explores the concepts of free energy, enthalpy, and entropy, distinguishing between endergonic and exergonic reactions and explaining redox potential, energy-rich compounds, and the significance of ATP and cyclic AMP. Carbohydrate metabolism is detailed with pathways such as glycolysis, the citric acid cycle, HMP shunt, gluconeogenesis, and glycogen metabolism, along with clinical conditions like G6PD deficiency, glycogen storage diseases, and diabetes mellitus. The discussion on biological oxidation emphasizes the electron transport chain, oxidative phosphorylation, substrate-level phosphorylation, and the effects of inhibitors and uncouplers. Lipid metabolism is explained through the oxidation of fatty acids, ketone body metabolism, de novo fatty acid synthesis, and the biological significance of cholesterol, including its conversion into bile acids, steroid hormones, and vitamin D, with attention to disorders such as hypercholesterolemia, atherosclerosis, fatty liver, and obesity. Amino acid metabolism covers transamination, deamination, decarboxylation, the urea cycle, and its disorders; catabolism of phenylalanine and tyrosine and their related diseases such as phenylketonuria, albinism, alcaptonuria, and tyrosinemia. It also highlights the synthesis and significance of key biological molecules such as dopamine, serotonin, 5-HT, melatonin, noradrenaline, and adrenaline, alongside heme catabolism, hyperbilirubinemia, and jaundice. The section on nucleic acid metabolism and genetic information transfer describes the biosynthesis and catabolism of purine and pyrimidine nucleotides, clinical conditions like gout, the structural organization of DNA and RNA, DNA replication, transcription, the genetic code, translation, and inhibitors of protein synthesis. The chapter on enzymes introduces their properties, nomenclature, and IUB classification, followed by enzyme kinetics explained through Michaelis-Menten and Lineweaver-Burke plots. It also addresses enzyme inhibition with examples, regulatory mechanisms including induction, repression, and allosteric regulation, and their therapeutic and diagnostic applications. Isoenzymes are discussed in medical contexts while

the section on coenzymes explains their structures and biochemical functions Altogether this textbook integrates biochemical principles with clinical correlations making it an essential guide for pharmacy medical and life science students bridging molecular understanding with disease mechanisms and therapeutic applications

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