

# Elements, Compounds, Mixtures, & Solutions

NAME \_\_\_\_\_ **TEACHER ANSWER KEY** \_\_\_\_\_ ROTATION \_\_\_\_\_

Part 1: Read the following information on elements, compounds, and mixtures. Fill in the blanks.

## Elements:

A pure substance containing only one kind of atom.



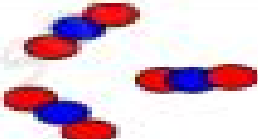
## Compounds:

The atoms are chemically combined in some way. Often (but not always), they come together to form groups of atoms called molecules.

## Mixtures:

Mixtures can be uniform (called homogeneous) and are known as solutions.

Draw a picture to represent the following:

Element	Compound	Mixture
		

Part 2: Classify each of the following as elements (E), compounds (C), or Mixtures (M). Write the letter X if it is none of these.

<u>E</u> Diamond (C)	<u>C</u> Sugar( $C_6H_{12}O_6$ )	<u>M</u> Milk	<u>E</u> Iron (Fe)
<u>M</u> Air	<u>C</u> Sulfuric Acid ( $H_2SO_4$ )	<u>M</u> Gasoline	<u>X</u> Electricity

# Answer Key For Solutions And Other Mixtures

**Peter Ellis, Derek McMonagle**



## **Answer Key For Solutions And Other Mixtures:**

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