

Name _____ Date _____ Class _____

CHAPTER

15

STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

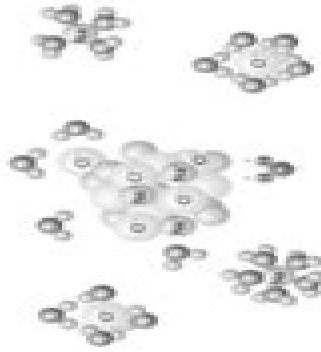
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
miscible	insoluble	solvent	solute

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?
Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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Section 15.1 continued

16. An sodium chloride dissolves in water. What happens to the sodium and chloride ions? The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the orientation of the water molecules around the sodium ions and chloride ions. Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

19. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

- true** 20. The overall energy change that occurs when a solution forms is called the heat of solution.
- false** 21. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
- false** 22. Solvation continues as long as the solubility rate is less than the crystallization rate.
- true** 23. In a saturated solution, solvation and crystallization are in equilibrium.
- true** 24. Additional water can be dissolved in an unsaturated solution.
- true** 25. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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E Durkheim



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formulas it is about understanding how molecules interact change and influence each other under different conditions With that in mind this book emphasizes the logic and patterns behind organic reactions making it easier for readers to apply concepts across a variety of scenarios Each chapter of this book builds upon foundational knowledge ensuring a progressive learning experience From nucleophilic substitutions to pericyclic reactions and from oxidation reduction mechanisms to named reactions we cover both fundamental and advanced topics to cater to students at all levels Real world examples have been integrated throughout the chapters to show how organic reactions play essential roles in pharmaceuticals biochemistry agriculture and environmental science This approach bridges the gap between theory and practical applications helping readers appreciate the relevance of organic chemistry in daily life *Prentice Hall Chemistry ,2000 Life* William K. Purves,2004 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR **Fundamental Concepts and Skills for Nursing - E-Book** Susan C. deWit,Patricia A. Williams,2013-01-24 NEW Rationales for NCLEX review questions at the end of each chapter help you understand why your choices were correct or incorrect NEW Full text reviews by experts in the field offer consistency and ease understanding as you progress through the book NEW Evolve margin icons denote supplemental material for students on Evolve NEW Evidence Based Practice margin icons point out the most current and evidence based information NEW In depth discussion of the Quality and Safety Education for Nurses QSEN within the text provides the knowledge skills and attitudes necessary to continuously improve the quality and safety of the healthcare systems **Fundamental Concepts and Skills for Nursing** Susan C. deWit,Patricia A. Williams, RN, MSN, CCRN,2013-01-18 Part of the popular LPN Threads series this market leading text features an easy to follow writing style and organization to teach you the concepts and skills you need to know to practice nursing in a variety of care settings This new edition offers in depth discussion of QSEN competencies rationales for the NCLEX PN review questions and new icons to point out the most current evidence based information Standard LPN Threads features include helpful characteristics such as full color design key terms numbered objectives key points critical thinking questions critical thinking activities glossary and references Easy to follow reading level and text organization presents information from simple to most complex Think Critically boxes help you synthesize information and apply concepts beyond the scope of the chapter Cultural Considerations related to biocultural variations as well as health promotion for specific ethnic groups demonstrate how to provide culturally competent care Elder Care Points highlight changes that occur with aging and how they affect nursing care Home Care

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