

Another Balancing Equations Sheet! – Answers

Balance these equations!

Note to students: Whenever balancing an equation, it is acceptable to leave spaces blank instead of writing "1" – in chemistry, they mean the same thing.

- 1) $1 \text{ AlBr}_3 + 3 \text{ K} \rightarrow 3 \text{ KBr} + 1 \text{ Al}$
- 2) $1 \text{ FeO} + 1 \text{ PdF}_2 \rightarrow 1 \text{ FeF}_2 + 1 \text{ PdO}$
- 3) $1 \text{ P}_4 + 6 \text{ Br}_2 \rightarrow 4 \text{ PBr}_3$
- 4) $2 \text{ LiCl} + 1 \text{ Br}_2 \rightarrow 2 \text{ LiBr} + 1 \text{ Cl}_2$
- 5) $1 \text{ PbBr}_2 + 2 \text{ HCl} \rightarrow 2 \text{ HBr} + 1 \text{ PbCl}_2$
- 6) $2 \text{ CoBr}_3 + 3 \text{ CaSO}_4 \rightarrow 3 \text{ CaBr}_2 + 1 \text{ Co}_2(\text{SO}_4)_3$
- 7) $2 \text{ Na}_3\text{P} + 3 \text{ CaF}_2 \rightarrow 6 \text{ NaF} + 1 \text{ Ca}_3\text{P}_2$
- 8) $2 \text{ Mn} + 6 \text{ HI} \rightarrow 3 \text{ H}_2 + 2 \text{ MnI}_3$
- 9) $1 \text{ Li}_3\text{PO}_4 + 3 \text{ NaBr} \rightarrow 1 \text{ Na}_3\text{PO}_4 + 3 \text{ LiBr}$
- 10) $1 \text{ CaF}_2 + 1 \text{ Li}_2\text{SO}_4 \rightarrow 1 \text{ CaSO}_4 + 2 \text{ LiF}$
- 11) $2 \text{ HBr} + 1 \text{ Mg}(\text{OH})_2 \rightarrow 1 \text{ MgBr}_2 + 2 \text{ H}_2\text{O}$
- 12) $2 \text{ LiNO}_3 + 1 \text{ CaBr}_2 \rightarrow 1 \text{ Ca}(\text{NO}_3)_2 + 2 \text{ LiBr}$
- 13) $1 \text{ AgNO}_3 + 1 \text{ Li} \rightarrow 1 \text{ LiNO}_3 + 1 \text{ Ag}$
- 14) $1 \text{ Si}(\text{OH})_4 + 4 \text{ NaBr} \rightarrow 1 \text{ SiBr}_4 + 4 \text{ NaOH}$
- 15) $2 \text{ NaCN} + 1 \text{ CuCO}_3 \rightarrow 1 \text{ Na}_2\text{CO}_3 + 1 \text{ Cu}(\text{CN})_2$

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Y Pai



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