

General Instructions :

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises **15** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **three** sections – **A, B** and **C**.
- (iii) **Section A** – Questions No. **1** to **7** are short answer type questions. Each question carries **2** marks.
- (iv) **Section B** – Questions No. **8** to **13** are also short answer type questions. Each question carries **3** marks.
- (v) **Section C** – Questions No. **14** and **15** are case-based questions. Each question carries **4** marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

- 1. (a) Write the molecular formula of the following carbon compounds :
 - (i) Methane
 - (ii) Propane
- (b) Carbon compounds have low melting and boiling points. Why ? 2
- 2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive ? 2
- 3. (a) Which of the following flowers will have higher possibility of self-pollination ?
Mustard, Papaya, Watermelon, Hibiscus
- (b) List the two reproductive parts of a bisexual flower. 2
- 4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2



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TABLE OF CONTENT Solved Paper 2019 December Solved Paper 2019 July Solved Paper 2018 December Solved Paper 2016 September Child Development and Pedagogy English Language and Pedagogy Hindi Bhasha evm Shiksha Shastra Mathematical Pedagogy Environmental Studies and Pedagogy Practice Sets 1 2 **2015-2016 Assessment of the Army Research Laboratory** National Academies of Sciences, Engineering, and Medicine,Division on Engineering and Physical Sciences,Laboratory Assessments Board,Army Research Laboratory Technical Assessment Board,2017-05-01 The National

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Table of Contents 2014physical Science Question Paper

1. Understanding the eBook 2014physical Science Question Paper
 - The Rise of Digital Reading 2014physical Science Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 2014physical Science Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2014physical Science Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2014physical Science Question Paper
 - Personalized Recommendations
 - 2014physical Science Question Paper User Reviews and Ratings
 - 2014physical Science Question Paper and Bestseller Lists
5. Accessing 2014physical Science Question Paper Free and Paid eBooks
 - 2014physical Science Question Paper Public Domain eBooks

- 2014physical Science Question Paper eBook Subscription Services
- 2014physical Science Question Paper Budget-Friendly Options
- 6. Navigating 2014physical Science Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - 2014physical Science Question Paper Compatibility with Devices
 - 2014physical Science Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2014physical Science Question Paper
 - Highlighting and Note-Taking 2014physical Science Question Paper
 - Interactive Elements 2014physical Science Question Paper
- 8. Staying Engaged with 2014physical Science Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2014physical Science Question Paper
- 9. Balancing eBooks and Physical Books 2014physical Science Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2014physical Science Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 2014physical Science Question Paper
 - Setting Reading Goals 2014physical Science Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2014physical Science Question Paper
 - Fact-Checking eBook Content of 2014physical Science Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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