

34. The maximum number of covalent bonds formed by nitrogen is

- A. 1.
- B. 2.
- C. 3.
- D. 4.

35. The IUPAC name of the compound $\text{CH}_3\text{CH}(\text{CH}_3)\text{CHCH}_2$ is

- A. 2 - methyl but - 1 - ene.
- B. 2 - methyl but - 2 - ene.
- C. 3 - methyl but - 1 - ene.
- D. 3 - methyl but - 2 - ene.

36. Ionization energy increases across the period in the periodic table because

- A. atomic number increases.
- B. effective nuclear charge increases.
- C. mass number decreases.
- D. screening effect decreases.

37. Which of the following properties indicate that an element is a metal. It

- I. reacts with oxygen to form an acidic oxide
- II. forms ionic chlorides.
- III. has variable oxidation states.
- IV. displaces hydrogen from dilute HCl.

- A. I and III only
- B. I and II only
- C. II and IV only
- D. I, II, III and IV

38. The electron configuration of carbon atom in its excited state is

- A. $1s^2 2s^2 2p_x^1 2p_y^0$.
- B. $1s^2 2s^2 2p_x^1 2p_y^1$.
- C. $1s^2 2s^1 2p_x^1 2p_y^1 2p_z^1$.
- D. $1s^1 2s^2 2p_x^1 2p_y^1 2p_z^1$.

39. An oxide has the following properties. It

- I. is a white powder.
- II. reacts with HCl.
- III. reacts with NaOH.
- IV. is insoluble in water.

The oxide is

- A. alkaline.
- B. amphoteric.
- C. acidic.
- D. neutral.

40. Which of the following statements about atoms of a metal is correct? They

- A. readily accept electrons.
- B. are soft.
- C. are held together by covalent bond.
- D. are held together by a sea of electron cloud.

41. Which of the following sketches is a graphical representation of Boyle's law?



42. The atom with the electron configuration $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4$ is in

- A. period 4, p- block.
- B. period 3, p- block.
- C. period 4, d- block.
- D. period 3, d- block.

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